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# Optimization of Coagulation Parameters forTurbidity Removal Using Box-Behnken model

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## ABSTRACT

This study aims to determine the efficacy of using calcium hydroxide Ca(OH)2 in coagulation process to treat urban liquid effluent from wastewater purification plant. The response surface methodology was used. The effects and interactions between three key process parameters was enhanced applying Box-Behnken design. Factors studied were: coagulant dosage (g/L), coagulation speed (rpm), and coagulation time (min). The treatment efficiency was determined by the turbidity removal rate and the final pH of the treated wastewater. The statistical soundness of the generated model was determined using analysis of variance. The optimal model for determining the relationship between the variables is a second-order quadratic model (R2>98 %). The best turbidity reduction (91.3 %) was obtained using a 0.50 g/L Ca(OH)2 coagulant and 130.1 rpm for 5 minutes. Under these conditions, the ultimate pH of the effluent reached 8.

Keywords: Process, Coagulation, Optimization, Response Surface Methodology, Box-Behnken.

# 1. INTRODUCTION

The optimization of water treatment processes is of great importance because it improves efficiency and reduces the cost of treatment [1, 2]. One of the more modern methodologies for process optimization is Response Surface Methodology (RSM). It involves the design of experiments, analysis, and partial regressionfit modeling of experimental parameters [3,4]. The approach may combine multiple factors at once and reveal reciprocal interactions in a process' yield; it also minimizes the number of experimental tests necessary to achieve statistically acceptable results.

The current study adjusted the process variables previously known to affect the water coagulation process [5, 6]. RSM was used to assess the impacts of three variables (Lime dosage, coagulation speed, and coagulation time) and their effects on turbidity removal and the final pH of wastewater treated in order to eliminate the colloidal suspended matter responsible for the turbidity of the water and to achieve a final pH in the range of 6–8 to avoid a post-adjustment of pH due to the use of calcium hydroxide.

## 1.1. Related works and contributions

The RSM methodology is one of the more recent ways of multiple response optimizations that is being applied in industries. It is commonly used because to its low cost and low test requirement. RMS has been used to optimize different processes for wastewater treatment.

Khettaf et al. [7] used the coagulation-flocculation process to remove organic matter from surface water used for drinking water production. The variables influencing the responses are the initial pH of the water and the concentrations of coagulant and flocculant used.RSM results reveal that the optimal conditions were an initial pH of 6.9, using 0.133 g/L

and 0.06 g/L of coagulant and flocculant, respectively. The final pH obtained was 6.78 and the abatment reached 56% and 59% in terms of COD and UV-254 respectively.

In another work by Louhichi et al. [8], similar to Khettaf et al. [7], the wastewater from the vegetable oil refinery was also treated by coagulation--flocculation. This study also used RSM but Box--Behnken (BBD) as an optimization design. The same variable factors were adopted. The optimum was acquired at an initial pH of 9.23 using 2400 mg/L and 60.05 mg/L of coagulant and flocculant, respectively. The treatment eliminated 99% of the

COD and 100% of the turbidity.

Gökçek, et al. [9] optimized the coagulation process for COD, SS, and turbidity removal from slaughterhouse wastewater. The Box Behnken design was used. The various experimental factors were the alum coagulant dosage, the coagulation speed and the settling time. Optimization gave maximum removal (75,90, and 91%), respectively using 1g/L of alum, stirring at a speed of 150 rpm and allowing the sample to settle for 10 minutes. RSM has also been used in other studies to optimize wastewater treatment process. Table 1 presents an overview of the work carried out above.

Table 1.	Overview	of water and	wastewater	treatment	studies	carried	out with RSM
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Type of wastewater	Process	Design	Runs	Variables	responses	Reference
Surface water		CCD	29	X1=coagulant concentration X2= flocculant concentration X3= initial pH	Y1= COD removal Y2= absorbance Y3= final pH	[7]
Vegetable- oil refinery wastewater	Coagula tion	Box Behnken	29	X1=coagulant concentration X2= flocculant concentration X3= initial pH	Y1= turbidity removal Y2= COD removal	[8]
Slaughterhou sewastewater	Floccula tion	Box Behnken	28	X1=Coagulant concentration X2= coagulation speed X3= Settling time	Y1= COD removal Y2=SS removal Y3= Turbidity removal	[9]
Petroleum wastewater		CCD	13	X1= initial pH X2= coagulant concentration	y1= final pH Y2= COD removal Y3= turbidity removal Y4= TDS removal Y5= color removal	[10]
Palm oil mill wastewater	Electro-	Box Behnken	16	X1=time X2= Voltage X3= NaCl concentration	Y1= COD removal Y2=TSS removal Y3= TDS removal	[11]
Coagula Pharmaceut ical wastewat er		CCD	30	<ul> <li>x1= initial pH</li> <li>x2= Cefazolin concentration</li> <li>x3= Currentdensity</li> <li>x4= Electrolysis time</li> </ul>	Y1= Cefazolin removal	[12]

## 2. METHODOLOGY

## 2.1. Origin And Characteristic Of The Effluent

The studied effluent was sampled from the wastewater purification plant, located in the town of Sidi Bel Abbes (western Algeria). The effluent was collected from the clarification tank before any chemicaltreatment. The wastewater is characterized by a pH of 7,5, a turbidity of 489 NTU, and a dissolved salt levelof 1253 g/L.

# 2.2. Coagulation-Flocculation Tests And Experimental Design

Tests of coagulation flocculation were realized in a

flocculator device (Jar Test AOUA/UTC). As a coagulant, calcium hydroxide  $Ca(OH)_2$  was used. The volume of wastewater used is 400 mL per test.

To optimize the variables that affect the efficiency ofthe coagulation process, RSM was employed. The statistical software Design Expert 13.0 was utilized for the experiment design and data analysis .The Box- Behnken (BBD) design was used to optimize the turbidity removal and the final pH of the wastewater treated by coagulation method.

Three factors were varied: coagulant dosage (A), coagulation speed (B), and coagulation time (C). The coded levels and ranges of each factor are shown in Table 2. The responses of the design were: the turbidity removal (Y1) and the final pH (Y2). 15 experimental runs were performed.

Table 2. Experin	nental and	coded	values
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Factors	Symbol	Coded levels			
Faciois	Symbol	-1	0	+1	
coagulant dosage (g/L)	A	0.5	2	3.5	
Coagulation speed (rpm)	В	100	150	200	
Coagulation time (min)	С	3	5	7	

The model's fitness was assessed using a significance test and an analysis of variance. A quadratic equation model, expressed in Equation (1), was used to optimize the coagulation process.

$$Y = \beta_0 + \sum_{i=1}^{k} \beta_i X_i + \sum_{i=1}^{k} \beta_i X_i^2 + \sum_{i=1}^{k-1} \sum_{j=2}^{k} \beta_{ij} X_i X_j$$

Where Y is the response,  $X_i$  represents the input factor,  $\beta_0$  is the intercept, the linear influence of theinput factor  $X_i$  is denoted by  $\beta i$ , -the interaction effect between  $X_i$  and  $X_j$  is denoted by  $\beta i j$ , the quadratic effectof Xi is  $\beta_{ii}$ .

The turbidity removal efficiency (%) is calculated according to Equation (2):

Turbidity removal, (%) =  $[(1-(T_f)/T_i)]*100$ 

T<sub>i</sub> is the initial turbidity

(2)

T<sub>f</sub> is the final turbidity

#### 3. RESULTS AND DISCUSSION

The relationships between turbidity removal efficiency, the final pH of treated wastewater, and the factors (coagulant dosage, coagulant speed, and coagulant time) were evaluated. Table 3 shows the operating parameters, results of the experiments that were conducted and predicted data.

#### 3.1. Analysis Of Variances (ANOVA)

The results in Table 3 showed that the experimental data fit the model perfectly. Multiple regression analysiswas used to analyze the experimental data. The second order polynomial Equations (3) and (4) were used to relate the responses values and the coded factors. The synergistic effect is indicated by the positive sign in front of the parameters, and the antagonistic effect is indicated by the negative sign.

- $$\begin{split} Y1 &= 98.4467 \ \text{-}0.45375\text{A} \ \text{-}1.72125\text{B} \ \text{+}1.6275\text{C} \\ &\quad \text{-}1.3475\text{AB} \ \text{+}2.22\text{AC} \ \text{-}0.875\text{BC} \ \text{-}8.69958\text{A}^2 \\ &\quad \text{+}6.16042\text{B}^2 \ \text{-}8.06208\text{C}^2 \end{split} \tag{3}$$
- $Y2 = 9.09333 + 0.93875A + 0.6975B 0.38375C + 0.7275AB 0.33AC + 0.5375BC 0.374167A^{2} + 1.37333B^{2} + 0.845833C^{2}$ (4)

For each response, the BBD design gave statistical parameter data. The variance analysis (ANOVA) was used to generate these statistical data, which were then used to determine the optimization's significance.

Test	A : coagula	B: Coagu Iation	C: Coagulation time (min)	Y1 Turbidity re	: moval (%)	Y2: Final pH		
	nt dosage (g/L)	speed (rpm)	, í	Actual value	Preditcted value	Actual value	Preditcted value	
1	0.5	200	5	96.24	95.99	96.24	95.99	
2	0.5	150	7	81.61	81.42	81.61	81.42	
3	2	200	3	95.26	94.20	95.26	94.20	
4	0.5	100	5	97.86	95.99	97.86	95.99	
5	3.5	150	3	77.32	77.51	77.32	77.51	
6	3.5	100	5	98.27	98.52	98.27	98.52	
7	2	150	5	97.99	98.45	97.99	98.45	

Table 3. Experimental and predicted data (BBD design)

8	2	150	5	98.62	98.45	98.62	98.45
9	3.5	200	5	91.26	92.14	91.26	92.14
10	2	100	7	99.58	100.64	99.58	100.64
11	2	200	7	95.26	95.70	95.26	95.70
12	2	150	5	98.73	98.45	98.73	98.45
13	3.5	150	7	86.52	85.20	86.52	85.20
14	2	100	3	96.08	95.64	96.08	95.64
15	0.5	150	3	81.29	82.61	81.29	82.61

The parameters such as F-value, P-value,  $R^2$  and  $R^2$  adjusted were determined to assess the model's effectiveness. The Model F-value of 41.59 and 70.05 shown in Tables 4 and 5 suggest that the model is significant. P-values of the model terms less than  $5.10^{-2}$  are considered significant and confirm factor-response interactions. Based on this, the factors B and C and the interaction terms

 $A^2$ ,  $B^2$ , and  $C^2$  had significant individual and quadratic effects on the removal of turbidity. Also, the parameters A and C had significant interactive effects between them. Analysis revealed that all three factors had significant individual (A, B, C), quadratic ( $A^2$ ,  $B^2$ ,  $C^2$ ), and interactive (AB, AC, BC) effects on the final pH value.

Table 4. Fit statistics for the respon	se Y1 (turbidityremoval)
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	Sumof	Df	Mean	F-value	p-value
Source	Squares		Square		
Model	761.13	9	84.57	41.59	0.0004
A-Coagulantdosage	1.65	1	1.65	0.81	0.4094
B-Coagulationspeed	23.70	1	23.70	11.66	0.0190
C-Coagulationtime	21.19	1	21.19	10.42	0.0233
AB	7.26	1	7.26	3.57	0.1174
AC	19.71	1	19.71	9.70	0.0264
BC	3.06	1	3,06	1,51	0,2743
A²	279.44	1	279,44	137,43	< 0.0001
B <sup>2</sup>	140.13	1	140,13	68,92	0,0004
C <sup>2</sup>	239.99	1	239,99	118,03	0,0001
Lack of Fit	9.85	3	3.28	20.59	0.0467
Pure Error	0.3189	2	0.1594	/	/

**Table 5.** Fit statistics for the response Y2 (final pH)

Source	Sum of Squares	df	Mean Square	F-value	p-value
Model	25.89	9	2.88	70.05	0.0001
A- Coagulantdosage	7.05	1	7.05	171.71	< 0.0001
B- Coagulationspeed	3.89	1	3.89	94.79	0.0002
C- Coagulationtime	1.18	1	1.18	28.69	0.0030
AB	2.12	1	2.12	51.56	0,0008
AC	0.4356	1	0.4356	10.61	0.0225
BC	1.16	1	1.16	28.15	0.0032
A²	0.5169	1	0.5169	12.59	0.0164

B²	6.96	1	6.96	169.61	< 0.0001
C²	2.64	1	2.64	64.34	0.0005
Lack of Fit	0.1832	3	0.0611	5.54	0.1568
Pure Error	0.0221	2	0.0110		

Figure 1 shows the relationship between experimental (actual) and predicted model values. We can observe that the predicted values of the model follow the experimental values well because they are near the diagonal axis. The values of the coefficient of determination  $R^2$  of 0.98 and 0.99 (table 6) obtained for the removal of turbidity (Y1) and for the final pH of wastewater (Y2), respectively, revealed a good fit of themodel.



**Figure 1** predicted vs. actual values: (a) of turbidity removal (Y1), (b) of final pH (Y2)

Referring to table 6, the predicted  $R^2$  values for turbidity removal and the final pH are in the order of 0.7948 and 0.8857, which is in agreement with the adjusted  $R^2$  values of 0.9631 and 0.9780, respectively. The difference is less than 0.2, which indicates that the model adopted is significant.

The correct precision value must be greater than 4. The correct precision values for the rate of turbidity removal and the final pH are 20.08 and 23.47, respectively. Therefore, both models are accepted.

Parameter	Turbidity removal(%)	Final pH
Standard deviation	1.43	0.2026
Mean	92.79	10.08
R²	0.9868	0.9921
Adjusted-R <sup>2</sup>	0.9631	0.9780
Predicted-R <sup>2</sup>	0.7948	0.8857
Adequate precision	20.0855	23.4745

 Table 6. Statistical parameter values

#### 3.2. Optimization Of Parameters

To obtain maximum turbidity removal under

optimal conditions, an optimization analysis was performed. The objectives of this analysis and its limitations are presented in Table 7. Prior to the analysis, aims were specified to find the optimized process conditions. The final pH of the wastewater was chosen to be in range 6 and 8. It is known that calcium hydroxide in its coagulant form induces very alkaline final pH. So this range was chosen on the basis of obtaining a final pH of the treated water close to neutral.

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Parameter	Aim	Minimum level	Maximum level
A- Coagulant dosage	Minimum	0,5	3,5
B- Coagulation speed	In range	100	200
C- Coagulation time	In range	3	7
Turbidity removal	Maximize	50	100
Final pH	In range	6	8

Table 7.	Objectives	and ranges	of optimization
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According to the objectives of the study, the selected optimal conditions was, coagulant dosage of 0.5 g/L, coagulation speed of 130.1 rpm and coagulation time of 5 min. At these optimal conditions, the predicted removal of turbidity was

91.3% and the final pH was of 8. The desirability of these optimal conditions is 0.908. The 3D graph of this desirability is shown in Figure 2.



Figure 2 Desirability surface for optimal conditions Figures 3 and 4 show 3D surface diagrams for the turbidity removal and the final pH, respectively, using 0.5 g of coagulant.

The Figure 3 shows a minor interaction between the agitation speed and the reduction of turbidity; changing its value along its axis has no meaningful effect on the elimination of turbidity in the wastewater. While a considerable interaction between coagulation time and turbidity reduction can be noted in the same figure, a maximum range of elimination is reached between 4 and 6 min.





Figure 3 : 3D-surface diagram for the turbidity removal

The Figure 4 shows that the interactions between time or speed of stirring and the final pH are weak;

a change in the value of time or speed of stirring does not significantly affect the variation of the final pH.



Figure 4 : 3D-surface diagram for the final pH value

## **4. CONCLUSION**

This study aimed at optimizing the operating conditions of the coagulation process applied to the pre- treatment of urban liquid effluent in the wastewater purification plant in the town of Sidi Bel Abbes (Algeria). The Box-Behnken Design was used to assess the effects of coagulant dosage (calcium hydroxide), coagulation speed, and coagulation time on the reduction of turbidity and the final pH of the wastewater.

The results show that the coagulation speed and time individually influence the reduction of turbidity, while the final pH value is influenced by all three factors. A second-order mathematical model was found to be well-adjusted to the experimental data.

The optimal conditions obtained were a  $Ca(OH)_2$ dosage of 0.5 g/L and a coagulation speed of 130.1 min for a time of 5 minutes. The optimum

condition makes it possible to obtain maximum performance from the coagulation process while minimizing the total cost of the treatment.

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#### NOMENCLATURE

DOI:

Acron ym	Meaning
ANOV A	Analysis Of Variance
UV	Ultra Violet
BBD	Box-Behnken Design
SS	Suspended Solids
RM	Response Surface
S	Methodology
COD	Chemical Oxygen Demand