

Effects of Silver Nanoparticles on Seed Germination and Seedling Growth of Radish (*Raphanus Sativus* L.)

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Abstract—The use of silver nanoparticles (AgNPs) in a wide range of consumer products has the furthered need to assess its impact on organisms. To date, there have been very few studies examining the effects of AgNPs on plants. This study investigated the impact of AgNPs to seed germination rate, seedling growth, peroxidase (POD) enzyme activity and chlorophyll content of a vegetable plant *Raphanus sativus*. Results showed that both AgNPs and AgNO₃ had no significant effect on seed germination of *R. sativus*. Root length and root fresh weight was significantly reduced when exposed to 40 mg/L AgNPs or AgNO₃. It showed that the toxicity of AgNPs on *R. sativus* seedling growth is similar to AgNO₃. 40 mg/L of AgNPs and AgNO₃ significantly increase the activities of POD in *R. sativus* root and leaf. *R. sativus* bioassays employed in our experiments provided valuable information concerning the effects of AgNPs on plants.

Keywords—*raphanus sativus*; silver nanoparticles; seed growth; peroxidase; toxicity.

I. INTRODUCTION

Engineered nanoparticles (ENPs) are defined as intentionally produced particles that have a characteristic dimension between 1 and 100 nm in at least one dimension. ENPs possess novel properties that are not shared by non-nanoscale particles with the same chemical composition. Nanotechnology promises to far exceed the impact of the Industrial Revolution and is projected to become a \$1 trillion market by 2015^[1]. Among these ENPs, silver nanoparticles (AgNPs) have a wide range of current and potential future applications, including chemical catalysts^[2], spectrally selective coatings for solar energy absorption^[3], surface-enhanced Raman scattering for imaging^[4], and in particular, antimicrobial sterilization^[5]. However, these effective and biocidal properties have the potential to adversely affect organisms in the environment. It is likely that measurable concentrations of AgNPs will find their way into the environment through product use and disposal, where they may exert adverse impacts on organisms^[6]. Given their expected fate and potential toxicity, it is becoming widely recognized that the environmental impacts of nanomaterials need to be understood^[6, 7].

However, our understanding of what AgNPs do to organisms or within natural environments is very limited, though the number of toxicological studies of AgNPs is rapidly increasing. Most studies to date have been conducted on bacteria, animal and human cells^[8, 9, 10, 11] and algae^[12, 13, 14]. It is reported that that oxidative stress may play a significant role in the NPs toxicity in nitrifying bacteria^[9] and human hepatoma cells^[10], but little is known about the influence in plants. To date, there is few report to study the possible activity changes of peroxidase (POD), one of antioxidant enzymes as reactive oxygen species (ROS) scavengers, in plants when they exposed to NPs.

There are few studies that investigate the impacts of AgNPs on plants. Stampoulis et al. (2009) reported that when 100 nm AgNPs at 500 and 100 mg/L resulted in 57% and 41% decreases in plant biomass and transpiration, respectively, as compared to controls or to plants exposed to bulk Ag^[14]. Kumari et al. (2009) investigate cytotoxic and genotoxic impacts of AgNPs on root tip cells of *Allium cepa*., the result showed that AgNPs could penetrate plant system and may impair stages of cell division causing chromatin bridge, stickiness, disturbed metaphase, multiple chromosomal breaks and cell disintegration^[15]. The treatment of silver nanoparticles enhanced peroxidase and catalase activity of *Bacopa monnieri* (Linn.) Wettst^[16]. These findings suggest that plants, as an important component of the ecosystems, need to be included when evaluating the overall toxicological impact of the nanoparticles in environment.

Radish (*Raphanus sativus* L.) is an economically important root vegetable crop produced throughout the world^[17]. It may be exposed to NPs via irrigation water contaminated with NPs and presents a threat to its growth. In this study, we sought to examine: (a) to ascertain the level of AgNPs required to inhibit the germination of *R. sativus* seeds, (b) to observe the growth of the seedlings under the treatment of AgNPs, (c) to evaluate the possible effect on the activity of POD of *R. sativus* induced by the AgNPs, and (d) compare the effect of AgNPs and AgNO₃ on the seedlings.

II. MATERIALS AND METHODS

A. AgNPs and AgNO₃ preparation

AgNPs (AGS-WM2000) solution (2000mg/L) was obtained from Shanghai HuZheng Nano Technology Co., Ltd. Energy filtering transmission electron microscopy (TEM) was used to examine the particle shape and size of the AgNPs. AgNO₃ was obtained from Shanghai Fine Chemistry Material Institute. All reagents, obtained from various commercial sources, were of analytical or higher grades.

B. Plant culture and treatment

Seeds were immersed in a 75% alcohol solution for 1 min to ensure surface sterility. Then they were soaked in DI-water, nanoparticle suspensions or AgNO₃ solution for about 2 h after been rinsed three times with DI-water. Three pieces of filter paper was put into each 120 × 15 mm Petri dish, and 8 ml of a test medium was added. Seeds were then transferred onto the filter papers, with 20 seeds per dish. Petri dishes were covered and sealed with tape, they were then placed in a growth chamber in random order where temperatures were 25±1 °C and 12 h light. 1, 10 and 40 mg/L AgNPs were prepared for testing the effect of AgNPs on the germination of *R. sativus* seed and seedling growth. Ag ions were used to compare the toxicities of AgNPs. The treatment with deionized (DI) water was comparison to compare the toxicities of AgNPs and Agions.

C. Determination of biomass and POD activity

At the end of the 4 days exposure, the root length was measured with calipers, the fresh weight was measured by analytical balance. POD activity was determined by the method of MacAdam (1992) and modification^[18]. The reaction mixture contained 3 ml reaction liquid (100 mM sodium phosphate buffer (pH 7.0), 28 μl of 2-methoxyphenol, 19 μl of hydrogen peroxide (30%) and 0.2 ml of enzyme extract or distilled water for control. The Chlorophyll content was measured by Chlorophyll Meter Model SPAD-502.

D. Statistical analysis

All errors are expressed as standard deviations (SD). Differences between treatments for the different measured variables were tested using one-way ANOVA (SPSS 13.0.1 for Windows), followed by Tukey HSD tests when differences significant at p<0.05 were found.

III. RESULTS AND DISCUSSION

A. Characterization of AgNPs

AgNPs used for *R. sativus* toxicity studies were characterized by TEM (JEM-100CX II), which showed these nanoparticles were spherical in shape (Fig. 1). The average diameter of particle was 15 nm. The pH value of AgNPs solution was 7.0±0.5, The AgNPs density was 1.07 g/ml (Table 1).

TABLE 1 CHARACTERISTICS OF SILVER ANOPARTICLES USED IN THIS STUDY

Index ^a	Performance
Appearance	Tawny
Concentration(Ag)	≥5%
PH value	7.0±0.5
Particle diameter	15nm
Conductivity	Passed
Density	1.07 g/ml
Purity	99.99%

^aThe data provided by producer.

B. Effects of AgNPs and AgNO₃ on the growth of *R. sativus* seedlings

The germination rate of *R. sativus* ses ranged from 90% to 97%, and there is no significant difference between control, AgNO₃ and AgNPs treatment (Fig. 2). The results indicated that AgNO₃, AgNPs do not have accurate toxic effect on *R. sativus* seed germination. This is consistent with the previous studies that nanoparticles had less effect on seed germination compare to seedling growth^[19,20]. Seed coat plays a very important role in protecting the embryo from harmful external factors. Seed coats can have selective permeability[21]. Pollutants, though having obviously inhibitory effect on root growth, may not affect germination as they cannot pass through seed coats. This may explain the reason that seed germination was not obviously effected by AgNPs in this study.

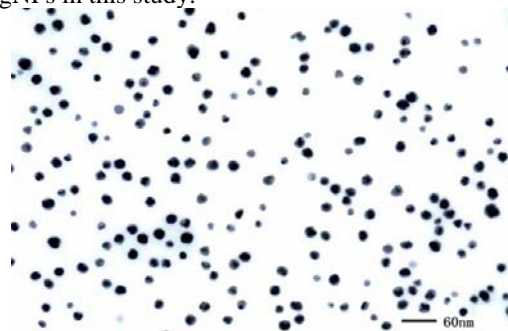


Figure 1. TEM images of 15 nm AgNPs.

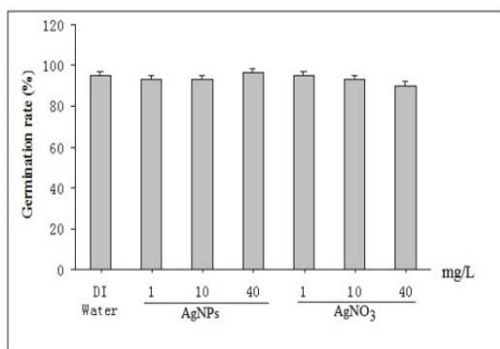


Figure 2. Effect of AgNPs and AgNO₃ on the germination rate of *R. sativus* seeds.

Both AgNO₃ and AgNPs affected the growth of *R. sativus* seedlings, Three kinds of AgNPs solution (1, 10 and 40 mg/L) had significant effect on root length, and significantly reduced root length (Fig.3). Similar effect was observed when AgNO₃ was used. The fresh weight of root was also affected by both AgNPs and AgNO₃. And there is significant difference between control and high concentration of AgNPs and AgNO₃ treatment. When 40 mg/L of AgNPs and AgNO₃ was applied, the fresh weight reduced more than 45% percent (Fig. 4). Recent studies of AgNPs have frequently attributed their toxicity to the release of dissolved silver^[12, 13, 5], and thus ionic silver is much more toxic than AgNPs at the same concentration. Oukarroum et al (2013) reported that the significant decrease of frond numbers of the aquatic plant *Lemna gibba* was dependent on AgNPs concentration^[22]. In this study, root length and root fresh weight was significantly reduced when exposed to 40 mg/L AgNPs or AgNO₃. It showed that the toxicity of AgNPs on *R. sativus* seedling growth is similar to AgNO₃.

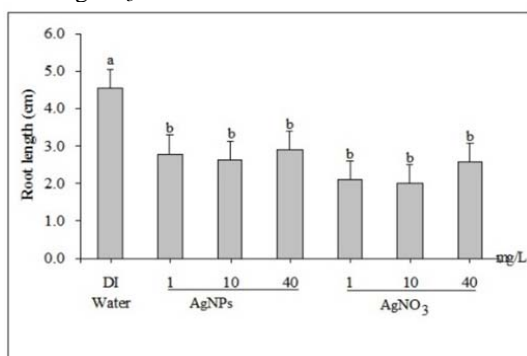


Figure 3. Effect of AgNPs and AgNO₃ on the root length of *R. sativus*. Different letters show significant differences (P<0.05).

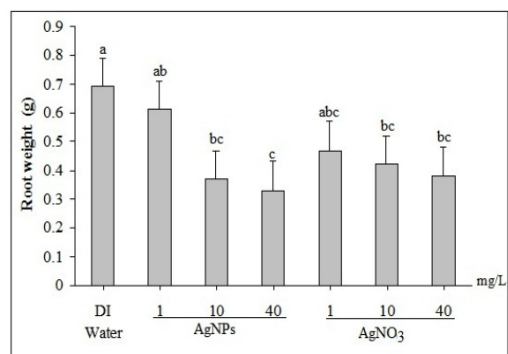


Figure 4. Effect of AgNPs and AgNO₃ on the root weight of *R. sativus*. Different letters show significant differences (P<0.05).

C. Effects of AgNPs and AgNO₃ on the activity of POD

POD activity, the protective enzyme for ROS stress, was measured, the results indicated that higher concentration of AgNO₃ (10 and 40 mg/L) significantly aroused the POD enzyme activity in roots, while lower dose of AgNO₃ (1 mg/L) had no significant effect when compared with the control (Fig. 5). AgNPs at 40 mg/L significantly enhanced the root POD activity, while 1 mg/L and 10 mg/L AgNPs had no significant effect when compared with the control. The POD activity in leaf indicated that both AgNO₃ and AgNPs at 1 mg/L and 10 mg/L had no significant effect when compared with control. The POD activity in leaf significantly increased with treatments of AgNO₃ and AgNPs at 40 mg/L. (Fig. 6). It was shown previously that ROS are generated during plant metabolism, especially in the plants exposed to environmental stresses, and they need to be scavenged for maintenance of normal growth. Much evidence has accumulated from various plant systems showing that environmental stresses alter the activities of enzymes involved in scavenging ROS^[23, 24, 25]. Among these enzymes, POD is one of the most important enzymes active in elimination of ROS. Previously studies showed that both dissolved silver and AgNPs can induce ROS production in bacteria and human hepatoma cells^[9, 10]. Peroxidase activity in the leaves of *B. monnieri* treated with AgNPs was significantly high and slightly higher than that of leaves treated with AgNO₃ from 20th day of exposure^[16]. In view of this, we investigated the activities of POD, one endogenous protective enzyme, to determine whether a general oxidative stress is also induced by AgNPs in plants. The results in this study demonstrated that high concentration of AgNPs and AgNO₃ (40 mg/L) significantly increase the activities of POD in *R. sativus* root and leaf, which is consistent with the seedling biomass response. The findings suggest that POD may take part in the process in which plants react against the AgNPs stress.

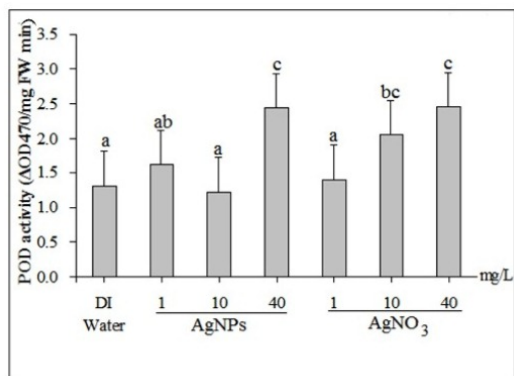


Figure 5. Effect of AgNPs and AgNO₃ on the POD activity of *R. sativus* root. Different letters show significant differences (P<0.05).

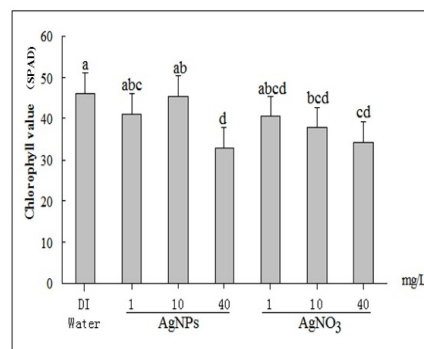


Figure 7. Effect of AgNPs and AgNO₃ on the contents of chlorophyll of *R. sativus* leaf. Different letters show significant differences (P<0.05).

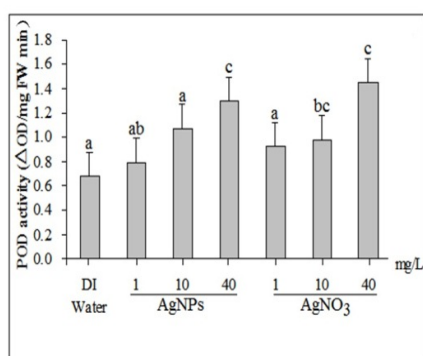


Figure 6. Effect of AgNPs and AgNO₃ on the POD activity of *R. sativus* leaf. Different letters show significant differences (P<0.05).

D. Effects of AgNPs and AgNO₃ on the chlorophyll content of leaf

Chlorophyll was the most important element in the photosynthesis. The chlorophyll contents of leaf reduced with the AgNPs and AgNO₃ treatments. The higher concentration of AgNO₃ (10 and 40 mg/L) significantly reduced the chlorophyll content in leaves, while lower dose of AgNO₃ (1 mg/L) had no significant effect when compared with the control (Fig. 7). The chlorophyll content in leaves significantly decreased at AgNPs at 40 mg/L.

IV. CONCLUSIONS

In this work, *R. sativus* bioassays employed in our experiments provided valuable information concerning the effects of AgNPs on plants. And our results indicate that exposure to AgNPs via irrigation water contaminated by AgNPs poses a threat to the yield and quality of vegetable plants in the environment. These results will help to further understand phytotoxicity of various nanomaterials, and are significant in terms of use and disposal of engineered nanoparticles. Future studies should be directed to phytotoxicity mechanisms of AgNPs, possible uptake and translocation of nanoparticles by plants and to examine whether the toxicity of AgNPs observed in this study also occurs in other plant species.

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REFERENCES

- [1] A. Nel, T. Xia, L. Mädler and N. Li: Science. 311(2006), p. 622-627
- [2] H.J. Zhai, D.W. Sun and H.S.Wang: J. Nanosci. Nanotechnol. 6(2006), p. 1968-1972
- [3] R.B.Rand, P. Peumans and S.R.Forrest: J. Appl. Physics. 96(2004), p.7519-7526
- [4] S. Yamamoto and H. Watarai: Langmuir. 22(2006), p. 6562-6569
- [5] S. Pal, Y.K.Tak and J.M.Song: Applied and Environ. Microbiol. (73) 2007, p.1712-1720
- [6] C.O. Robichaud, D.Tanzil, U. Weilenmann and M.R. Wiesner: Environ.Sci. Technol. 39 (2005), p. 8985-8994
- [7] M.R.Wiesner, G.V. Lowry, P.J. Alvarez, D. Dionysiou and P. Biswas: Environ.Sci. Technol. 40(2006), p. 4336-4345
- [8] J.R. Morones, J.L.Elechiguerra, A.Camacho, K. Holt, J.B. Kouri, J.T. Ramirez and M.J. Yacaman: Nanotechnology. 16(2005), p. 2346-2353
- [9] O. Choi and Z.Q.Hu: Environ. Sci. Technol. 42(2008), P.4583-4588
- [10] S. Kim, J.E.Choi, J. Choi, K.H. Chung, K. Park, J. Yi and D.Y. Ryu: Toxicol. in Vitro. 23(2009), p.1076-1084
- [12] A. Kahru and H.C. Dubourguier: Toxicology. 269 (2010), p.105-119

- [13] A.J. Miao, K.A. Schwehr, C. Xu, S.J. Zhang, Z.P. Luo, A. Quigg and P.H. Santschi: *Environ. Pollut.* 157 (2009), p. 3034-3041
- [14] Dewez and A.Oukarroum: *Toxicol. Environ. Chem.* 94(2012), p.1536-1546
- [15] D. Stampoulis, S.K. Sinha and J.C. White: *Environ. Sci. Technol.* 43 (2009), p. 9473-9479
- [16] M. Kumari, A. Mukherjee and N. Chandrasekaran: *Sci. Total Environ.* 407 (2009), p. 5243-5246
- [17] C. Krishnaraj, E.G. Jagan, R. Ramachandran, S.M. Abirami, N. Mohan and P.T. Kalaichelvan: *Process Biochem.* 47(2012): 651-658
- [18] L.Z. Wang and Q.W.He: *Chinese radish (Scientific and Technical Documentation Press, China 2005)*
- [19] J.W. MacAdam, R.E.Sharp and C.J. Nelson: *Plant Physiol.* 99(1992), p. 879-885
- [20] D. Lin and B. Xing: *Environ. Pollut.* 150(2007), p. 243-250
- [21] R.C. Monica and R. Cremonini: *Caryologia.* 62 (2009), p.161-165
- [22] M. Wierzbicka and J. Obidzinska: *Plant Sci.*137(1998), p.155-171
- [23] A. Oukarroum, L. Barhoumi, L. Pirastru and D. Dewed: *Environ. Toxicol. Chem.* 32(2013),p. 902-907
- [24] Y.Gueta-Dahan, Z.Yaniv, B.A. Zilinskas and G. Ben-Hayyim: *Environ. Pollut.* 159(1997), p. 1551-1559
- [25] L.Y.Yin, J.Q.Huang, W.M. Huang, D.H. Li, G.H. Wang and Y.D. Liu: *Toxicol.* 46(2005), p. 507-512
- [26] J.Q. Huang, H.F. Jiang, Y.Q. Zhou, Y. Lei, S.Y. Wang and B.S. Liao: *Int. J. Food Microbiol.* 130 (2009), p.17-21