

Process and Models of Decalcification of Bighead Carp Scale by Hydrochloric Acid

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Abstract. Collagen is one of the important biological materials. Fish scale contains rich collagen. Extracting collagen from fish scale not only reduces the biological pollution but also receives good economic benefit. In this paper, based on the experimental results of single factor test made in our previous research work, the response surface method was used to further optimize the technical parameters for the decalcification of bighead fish scale and establish the process model and kinetic model. The results showed that the maximum calcium removal rate reached 86.61% under the optimal conditions of hydrochloric acid concentration 5.0%, decalcification time 45 min and liquid/solid ratio 21.65 and the maximum decalcification velocity reached 0.4932g/(L min) under the optimal conditions of hydrochloric acid concentration 6.4%, decalcification time 10.16 min and liquid/solid ratio 10.23. The two models in the form of multiple regression were proved to be basically identical with the actual values and they can be used to predict the calcium removal rate and decalcification velocity under different operation conditions and the decalcification velocity corresponding to the calcium removal rate under same conditions, to adjust the operation parameters for gaining the expected calcium removal rate and decalcification velocity, and to control the production capacity of fish scale treatment and the quality and production output of collagen prepared by following collagen extraction from the decalcified fish scale.

Introduction

Bighead carp, also called *Aristichthys nobilis*, is one of the “four famous fishes” (black carp, grass carp, silver carp and bighead carp) in our country and one of the China's major freshwater aquaculture species. With the development of China's aquaculture, the fish products processing has gotten a rapid development. In the fish products processing, a large number of scraps, including fish scales, fish skin, fish innards and fishbone, etc., are produced. At present, this kind of resource has not effectively utilized and only a small part of it is used as raw material to prepare feed in our country. Most of the fish scraps are abandoned, which not only pollutes the environment, but also causes the waste of resources. Therefore, comprehensively utilizing the fish scales is very significant to protect environment and develop the domestic economy[1-4]. Fish scales contain plenty of protein, mainly collagen, which accounts for about 50% ~ 60% of total weight scales. Collagen is a kind of biological macromolecule material. It can supplement the nutrition for the skin layers, enhance the collagen activity of skin, have the functions of moistening skin, anti-aging, improving one's looks, eliminating wrinkle and raising hair etc. The fish scale contains a certain amount of hydroxyapatite. The collagen fibers are stucked with hydroxyapatite together, which will influence the process and quality of the collagen extraction of fish scale if hydroxyapatite or calcium is not removed as completely as possible before collagen extraction. The removal of hydroxyapatite is called decalcification process in which the calcium is dissolved from the apatite

lattice of hydroxyapatite. In recent years, some researchers have done the studies on fish scale decalcification with EDTA, hydrochloric acid and citric acid as extracting agents. Most of them focused on the optimization of the operation conditions. Decalcification of fish scale is a process with reaction between acid and salt. The process model reflects the relations between calcium removal rate (as target) and main operation parameters(as influence factors). The kinetic model reflects the relations between decalcification velocity (as target) and main operation parameters(as influence factors). These two models can predict and adjust the technical parameters to achieve the expected calcium removal rate and decalcification velocity under different operation conditions, which is conducive to controlling the production capacity of fish scale treatment and collagen quality. In this paper, response surface method was used to further optimize the operation parameters and establish the process and kinetic models of decalcification from bighead fish scale and validate the models with the actual values, based on the results of single factor test method in our previous research work[5].

Experimental

Raw materials

Bighead carp scales: collected from local market. The scales were cleaned with distilled water and dried for use .

Chemicals: calcium carbonate, complexoneII(EDTA), hydrochloric acid, chrome black T, hydroxylamine hydrochloride, ethanol, ammonium chloride and ammonia. All the above are analytically pure.

Standard Regression Equation of Calcium Concentration

5.00g anhydrous calcium carbonate, dried to constant weight at 110 °C, was accurately weighed and added into a breaker (100 mL) . Then a small amount of distilled water and hydrochloric acid solution with concentration of 1 mol/L were added into the breaker until the calcium carbonate was completely dissolved. After that, the solution was transferred to a 1000 ml volumetric flask and diluted to scale. The concentration of calcium ion was 2 mg/mL.

$$y = 0.02283x - 0.00439 \quad R = 0.9996(p < 0.0001) \quad (1)$$

0.00, 1.00, 2.00, 3.00, 4.00 and 6.00 mL of standard calcium solution above were respectively added in 6 conical flasks (250 mL) and diluted to 20 mL with distilled water. Then 15 mL of ammonia buffer (pH = 10) and 3 drops of chrome black T indicator were added into the flasks. The solution was measured by titration with 0.01 mol/L EDTA standard solution until the color of the solution was changed from purple to blue. The consumed volume of EDTA was recorded. This measurement was conducted for two times in parallel. After that, the calcium standard curve was drawn with consumed volume of EDTA as the abscissa and calcium ion concentration as the ordinate[6]. The standard regression equation (liner equation) was obtained according to the standard curve.

Measurement of Calcium Removal Rate

1g bighead fish scales was accurately weighed and added in a 50 mL conical flask, then the hydrochloric acid solution with certain concentration and certain volume was added in flask. The flask was placed in a oscillator and the fish scales were decalcified under oscillation. After a certain time, the liquid-solid mixture was filtered and the filtrate was transferred to a 50 mL volumetric flask. The conical flask and filter paper were washed for three times with total 10 mL of distilled water. The washing water was merged into the filtrate and then diluted to scale.

1.00 mL of the solution in the volumetric flask above was taken and transferred into a 250 mL conical flask and diluted with distilled water to 20 mL. Then 15 mL ammonia buffer(pH=10) and 3

drops of chrome black T indicator were added into the flask. The solution was measured by titration with 0.01 mol/L EDTA standard solution until the color of it was changed from purple to blue and the consumed volume of EDTA was recorded. The calcium ion concentration in the decalcified solution was calculated by equation (1) and then the calcium removal rate (y) was calculated using following equation.

$$y(\%) = \frac{c \times V_1 \times V_2}{V_3 \times m_1 \times x \times 1000} \times 100 \quad (2)$$

Where c is the calcium ion concentration calculated by equation (1), mg/mL; V_1 is the diluted volume of the decalcified solution for titration, mL; V_2 : is the constant volume of the decalcified solution, mL; V_3 is the volume of decalcified solution from the 50mL volumetric flask, mL; m_1 is the mass of bighead fish scale, g; x is the mass fraction of calcium in fish scale, 13.9% [5].

Measurement of Decalcification Velocity

The decalcification velocity of fish scale is defined as mass of decalcified calcium per unit volume and per unit time, g/(L min), which is calculated by following equation:

$$v = \frac{y \times m_1}{V \times \tau} \quad (\text{g}/(\text{L} \cdot \text{min})) \quad (3)$$

Where V is the volume of the hydrochloric acid solution, mL; τ is the time of decalcification, min.

Establishment of Models

The suitable operation parameters for decalcification of bighead fish scale were determined by single factor test: mass fraction of hydrochloric acid 5%, decalcification time 40 min and ratio of liquid/solid 20mL/g, under which, the calcium removal rate reached 84.87% [5]. Based on the results of single factor test, the response surface test (Box-Behnken) was carried out in this experiment. Hydrochloric acid, decalcification time and ratio of liquid/solid were taken as investigation factors and calcium removal rate and decalcification velocity were taken as targets. The design of three factors with three levels was listed in Tab.1.

According to the design above, 17 sets of test (including 5 sets of central group tests) were separately conducted when the calcium removal rate and decalcification velocity were taken as targets and the results were listed in Tab.2.

The results of variance analysis of regression model with calcium removal rate and decalcification velocity as targets were exhibited in Table 3 and Table 4, respectively.

$$Y = -29.3288 + 88.3125X_1 + 2.0921X_2 + 3.6904X_3 + 0.0588X_1X_2 + 0.1250X_1X_3 - 0.0118X_2X_3 - 74.50X_1^2 - 0.0203X_2^2 - 0.0731X_3^2 \quad (4)$$

$$Y = 0.6558 + 0.6450X_1 - 0.0124X_2 + 0.0265X_3 - 1.0000 \times 10^{-3}X_1X_2 - 2.0000 \times 10^{-3}X_1X_3 + 1.8250 \times 10^{-4}X_2X_3 - 0.5250X_1^2 + 6.4375 \times 10^{-5}X_2^2 + 6.9250 \times 10^{-4}X_3^2 \quad (5)$$

Tab.1 Test design of three factors/three levels for response surface method

factor	code	levels		
		-1.00	0.00	1.00
Hydrochloric acid concentration / (mass)%	X_1	4.00	5.00	6.00
Decalcification time /min	X_2	20.00	40.00	60.00
Ratio of liquid to solid/ mL/g	X_3	10.00	20.00	30.00

Tab.2 Results of response surface method

Order No.	X ₁	X ₂	X ₃	calcium removal rate		decalcification		velocity	
				Test	Model	Test	Model	relative	
				relative error		g/(L min)		%	
				%				%	
1	4.00	20.00	20.00	67.57	69.10	2.26	0.235	0.240	2.13
2	6.00	20.00	20.00	71.00	72.60	2.25	0.250	0.260	4.00
3	4.00	60.00	20.00	80.80	79.20	-1.98	0.094	0.085	-9.57
4	6.00	60.00	20.00	84.70	83.17	-1.81	0.098	0.089	-9.18
5	4.00	40.00	10.00	69.95	71.53	2.26	0.243	0.250	2.88
6	6.00	40.00	10.00	73.50	75.01	2.05	0.255	0.260	1.96
7	4.00	40.00	30.00	79.94	78.43	-1.89	0.093	0.085	-8.60
8	6.00	40.00	30.00	83.99	82.41	-1.88	0.097	0.089	-8.25
9	5.00	20.00	10.00	61.47	58.36	-5.06	0.42	0.41	-2.38
10	5.00	60.00	10.00	73.38	73.40	0.03	0.170	0.170	0.00
11	5.00	20.00	30.00	70.24	70.22	-0.03	0.163	0.160	-1.84
12	5.00	60.00	30.00	72.72	75.83	4.28	0.076	0.073	-3.95
13	5.00	40.00	20.00	84.90	84.90	0.00	0.150	0.150	0.00
14	5.00	40.00	20.00	84.90	84.90	0.00	0.150	0.150	0.00
15	5.00	40.00	20.00	84.90	84.90	0.00	0.150	0.150	0.00
16	5.00	40.00	20.00	84.90	84.90	0.00	0.150	0.150	0.00
17	5.00	40.00	20.00	84.90	84.90	0.00	0.150	0.150	0.00

Tab.3 Results of variance analysis of regression model (with calcium removal rate as target)

Source	Sum of square	Degree of freedom	Mean square	F value	Prob>F	Significance level
		DF	MS			
model	908.30	9	100.93	18.21	0.0005	significant
X ₁	27.86	1	27.86	5.03	0.0599	
X ₂	213.42	1	213.42	38.51	0.0004	
X ₃	102.17	1	102.17	18.44	0.0036	
X ₁ X ₂	0.055	1	0.055	9.965×10 ⁻³	0.9233	
X ₁ X ₃	0.063	1	0.063	0.011	0.9184	
X ₂ X ₃	22.23	1	22.23	4.01	0.0853	
X ₁ ²	2.34	1	2.34	0.42	0.5368	
X ₂ ²	278.82	1	278.82	50.31	0.0002	
X ₃ ²	224.99	1	224.99	40.60	0.0004	
Residual	38.79	7	5.54			
Lack of Fit	38.79	3	12.93	4.69		Not significant
Pure error	0.000	4	0.000			
Total variation	947.13	16				
R ²	0.9590					

It can be seen from Table 3 that the model F-value of 18.21 implies the model is significant. There is only a 0.05% chance that “a Model F-value” this large could occur due to noise. The values of “ Prob>F” less than 0.0500 indicate the model terms are significant, in this case, X₂, X₃, X₂X₃, X₂² and X₃² are significant model terms. The values of “ Prob>F” greater than 0.1000 indicate the model terms are not significant, in this case, X₁, X₁X₂, X₁X₃ and X₂X₃ are not significant model terms. The correlation coefficient of the model is R²=0.9590, which means that the model can well describe the test results and only 4.10% of total variation of response value can not be explained by the model. Furthermore, as shown in the 7th column of “relative error” of Table 2, the model values of calcium removal rate are basically close to the test values and their relative errors are ranged

from 0.00% to 5.06% (absolute values), indicating that the model is basically accurate to predict and adjust the process parameters for gaining the expected calcium removal rate. According to the “Prob>F” values of X_1 , X_2 , and X_3 in Table 3, the significance of the effects of the three factors on the calcium removal rate can be determined to be $X_2 > X_3 > X_1$. The technical parameters were further optimized to be hydrochloric acid concentration 5%, decalcification time 45 min and liquid/solid ratio 21.7 by the solution of the model (4) with the software. Under the optimal conditions, the maximum calcium removal rate was 86.61%.

It can be seen from Table 4 that the model F-value of 82.83 implies the model is significant. There is only a 0.01% chance that “a Model F-value” this large could occur due to noise. The values of “ Prob>F” less than 0.0500 indicate the model terms are significant, in this case, X_2 , X_3 , X_1^2 , X_2^2 and X_3^2 are significant model terms. The values of “ Prob>F” greater than 0.1000 indicate the model terms are not significant, in this case, X_1 , X_1X_2 , X_1X_3 and X_1^2 are not significant model terms. The model “ Prob>F” value less than 0.0001 indicate that the relations between the response value (decalcification velocity) and the three factors are extremely significant and the test results are reliable. The correlation coefficient of the model is $R^2=0.9907$, which means that the model can well describe the test results and only 0.93% of total variation of response value can not be explained by the model. Furthermore, as shown in the 10th column of “relative error” of Table 2, the model values of decalcification velocity are basically close to the test values and their relative errors are ranged from 0.00% to 9.57% (absolute values), indicating that the model is basically accurate to predict and adjust the process parameters for obtaining the expected decalcification velocity. According to the “Prob>F” values of X_1 , X_2 and X_3 in Table 4, the significance of the effects of the three factors on the decalcification velocity can be determined to be $X_2 > X_3 > X_1$. The technical parameters were further optimized to be hydrochloric acid concentration 6.4%, decalcification time 10.16 min and liquid/solid ratio 10.23 by the solution of the model (5) with the software. Under the optimal conditions, the maximum decalcification velocity was 0.4932 g/(L min).

Tab. 4 Results of variance analysis of regression model (with decalcification velocity as target)

Source	Sum of square	Degree of freedom DF	Mean square MS	F value	Prob>F	Significance level
model	0.12	9	0.014	82.83	<0.0001	significant
X_1	1.280×10^{-4}	1	1.280×10^{-4}	0.77	0.4080	
X_2	0.053	1	0.053	319.52	<0.0001	
X_3	0.058	1	0.058	351.76	<0.0001	
X_1X_2	1.600×10^{-5}	1	1.600×10^{-5}	0.097	0.7648	
X_1X_3	1.600×10^{-5}	1	1.600×10^{-5}	0.097	0.7648	
X_2X_3	5.329×10^{-3}	1	5.329×10^{-3}	32.24	0.008	
X_1^2	1.161×10^{-4}	1	1.161×10^{-4}	0.70	0.4297	
X_2^2	2.792×10^{-3}	1	2.792×10^{-3}	16.89	0.0045	
X_3^2	3.602×10^{-3}	1	3.602×10^{-3}	21.79	0.0023	
Residual	1.157×10^{-3}	7	1.653×10^{-4}			
Lack of Fit	1.157×10^{-3}	3	3.857×10^{-4}			Not significant
Pure error	0.000	4	0.000			
Total variation	0.12	16				
R^2	0.9907					

Conclusions

The technical parameters were further optimized by response surface method. The optimal parameters were determined to be hydrochloric acid concentration 5.0%, decalcification time 45 min and liquid/solid ratio 21.65, under which the maximum calcium removal rate reached 86.61% and hydrochloric acid concentration 6.4%, decalcification time 10.16 min and liquid/solid ratio 10.23 under which the maximum decalcification velocity reached 0.4932g/(L min).The models

were basically in agreement with the test values. They can predict the calcium removal rate and decalcification velocity under different conditions and decalcification velocity corresponding to the calcium removal rate under same conditions which can adjust the parameters for gaining the expected calcium removal rate and decalcification velocity and control the production capacity of scale treatment and the collagen quality by following extraction from the decalcified fish scale.

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References

- [1]Duan, R.,Zhang, J.J., Du, X.Q., Yao, X.C.,& Konno Kunihiko: Food Chemistry 112(2009),p. 702.
- [2]Kanokwan,Matmaaroh, Soottawat Benjakul, Thummanoon Prodprun, Angel B., Hideki Kishmura: Food Chemistry 129(2011), p.1179.
- [3]Dasong Liu, Li Liang, Joe M., Regenstein, Pen.: Food Chemistry 133(2012),p. 1441.
- [4]Jiangnan Zeng, Boquan Jiang, Linsheng Wei, Yafang Zhang:: Nanchang University (Engineering & Technology) 33(2011),p.338.
- [5] Ning Zhengxiang. Food composition analysis manual [M]. Beijing: China, Light Industry Press, 1998:579-583.