

## Electric coagulation method and treatment of the experimental study of landfill leachate

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**Abstract.** Garbage leachate high concentrations of organic matter is considered difficult to deal with wastewater, The complex and changeable water quality makes traditional processing method difficult to achieve better treatment effect. So the electric coagulation method processing technology application becomes inevitable trend. Using Fe as the anode and C as the cathode to treat leachate of a certain garbage in electric coagulation depth processing, and explore the electrolysis time, initial PH value, plate spacing and current density effect of the treatment effect. Results show that when the initial PH value was 6, plate spacing was 3 cm and current density was 9ma/cm<sup>2</sup>, electrolysis 60 min reached optimal effect, the COD removal rate was 84.5% and the TSS removal rate was 69.2%.

### 1. Introduction

With the continuous development of society and improvement of people's living standards, environmental pollution of landfill leachate come along[1].Currently the common methods of landfill leachate treatment are biological method, chemical precipitation method and wetland method, etc[2,3],these methods have better treatment effect,but there are also many relevant issues,even be re-contaminated environment[4-6].Therefore, landfill leachate treatment new technologies research and analysis, research and application of it has become relevant departments urgent to think of topics[7-9].Compared with the conventional treatment methods,electric coagulation method requires no additional agents,the reaction time is shorter,the sludge production is less and the treatment efficiency is higher.In recent years it has been widely used in various types of wastewater treatment,and is worthy of further study[10].

### 2. Experimental device and analytical method

#### 2.1 Water sample parameters and emissions standards

Each parameter content of water samples and the corresponding emission standards as shown in the following table:

Table 1 content of water samples and the emission standards

index	content (mg/L)	emission standard
PH	6.5	6~9
COD	324	100
TSS	150	70

## 2.2 Experimental apparatus and experimental methods

### 2.2.1 Experimental apparatus

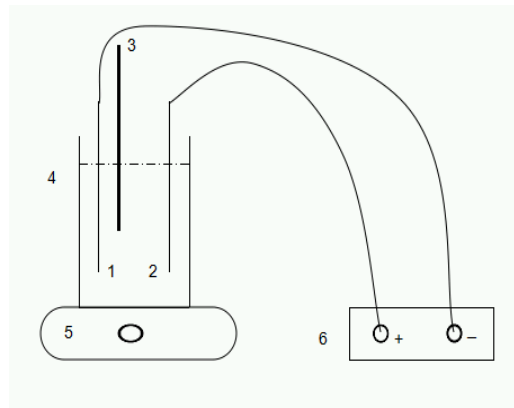


Fig. 2-1 Experimental apparatus

- 1—cathode    2—anode    3—thermometer  
4—electrolyzer (1000ml beaker)    5—magnetic stirrer  
6—DC stable constant current power supply

### 2.2.2 Experimental methods

Electric coagulation process also is known as electrolytic flocculation method[11].It refers to aluminum or iron as an electrode,In the role of DC,the anode is dissolution,produces flocculation ions,then through a series of hydrolysis, polymerization (and ferrous oxide) and other processes,the colloidal impurities and suspended impurities in the water take place flocculation and precipitation and separate[12].Electric coagulation process is mainly through electrolysis of metal anode (used as Fe, Al and other low-potential metal electrode), in the role of DC, The anode is dissolution,produce  $\text{Fe}^{2+}$ , $\text{Al}^{3+}$  ions, $\text{Fe}^{2+}$ , $\text{Al}^{3+}$  hydrolysis and polymerize in the solution,form various hydroxyl complexes,and then form polymer polynuclear hydroxyl complexes and hydroxide,these polynuclear hydroxyl complexes and hydroxide can make colloidal impurities and suspended impurities in wastewater flocculation precipitation and separated from water[13].

## 3. Results and analysis

### 3.1 The impact on the treatment effect of operation time

In conditions of iron anode,room temperature,The initial PH value of 6 and current density of 0.2A operations 60min,the impact on removal rate of electrolysis time is shown in Figure 3-1,COD and TSS removal rate increased with electrolysis time.After electrolysis 40min, increase in removal rate started slowing down,50min beginning to stabilize.The growth of electrolysis time was conducive to reaction fully,so removal rate increased. Therefore, we choose as a operation time of 60min.

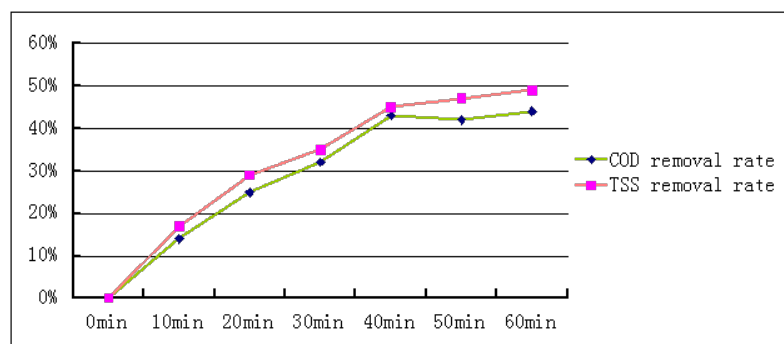


Fig. 3-1 The impact on removal rate of electrolysis time

### 3.2 The impact on the treatment effect of PH value of water samples

The experimental water samples PH solution, initial PH value are respectively 2,3,4,5,6,7, in conditions of electrolysis time of 60min and current density of  $11\text{mA}/\text{cm}^2$  experimental research the impact on the treatment effect of PH value of water samples, the results are shown in Figure 3-2.

Although it can demulsify under acid condition, but does not utilize production of flocs. According to experiments, under strong acid conditions, COD and TSS removal rate was not high, under weakly acid conditions (PH about 6), COD and TSS removal rate reached the highest. At this time COD removal rate was 78.7%, TSS removal rate was 61.0%.

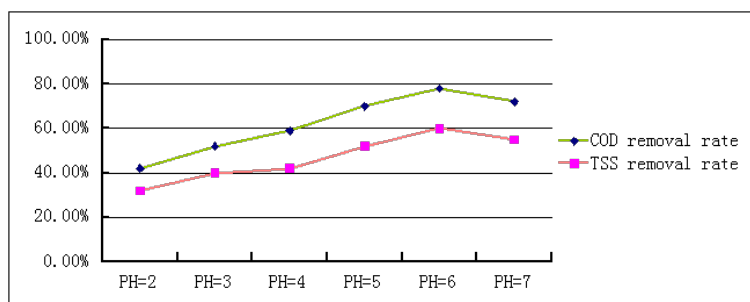


Fig. 3-2 The impact on the treatment effect of initial PH value

### 3.3 The impact on the treatment effect of current density

In conditions of iron as the anode and graphite as the cathode, adjusting the initial PH = 6, plate spacing of 3cm, current density of  $3\text{mA}/\text{cm}^2$ ,  $5\text{mA}/\text{cm}^2$ ,  $7\text{mA}/\text{cm}^2$ ,  $9\text{mA}/\text{cm}^2$ ,  $11\text{mA}/\text{cm}^2$ ,  $13\text{mA}/\text{cm}^2$ . After an hour electrolysis, COD and TSS removal rates were respectively 48.5%, 34.1%; 60.7%, 44.6%; 72.3%, 59.3%; 84.3%, 70.2%; 83.9%, 70.6%; 82.5%, 72.3%. Obviously, as the current density increased COD and TSS removal rates were on the rise, especially the magnitude of the rise was larger when the current density was less than  $7\text{mA}/\text{cm}^2$ . When the current density reached  $7\text{mA}/\text{cm}^2$ , rising rate began slowing down. When  $9\text{mA}/\text{cm}^2$ , COD and TSS removal rates were stable, then the COD removal rate was 84.5%, TSS removal rate was 69.2%.

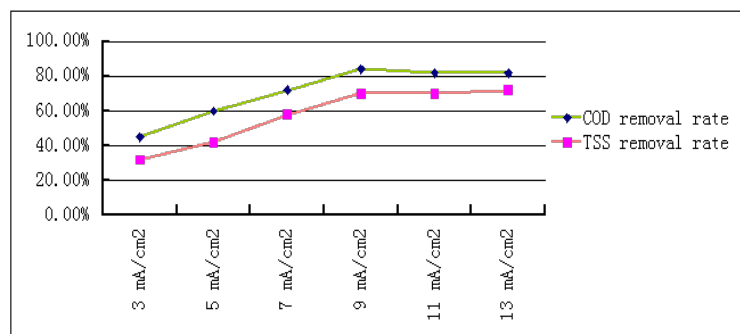


Fig. 3-3 The impact on the treatment effect of current density

### 3.4 The impact on the treatment effect of plate spacing

In conditions of iron as the anode and graphite as the cathode, adjusting the initial PH = 6, current density of  $9\text{mA}/\text{cm}^2$ , plate spacing of 2cm, 3cm, 5cm, 7cm. After an hour electrolysis, COD and TSS removal rates were respectively 60.2%, 48.1%; 82.4%, 64.3%; 57.8%, 52.6%; 50.5%, 38.2%. Obviously, small plate spacing was conducive to the removal of COD and TSS in the solution. However, when the plate spacing was less than 3cm removal rate was not high. The reason is plate spacing is too large or too small is not easy to mixing and diffusion of solution and the formation of flocs. Therefore I suggest to select 3cm as the plate spacing.

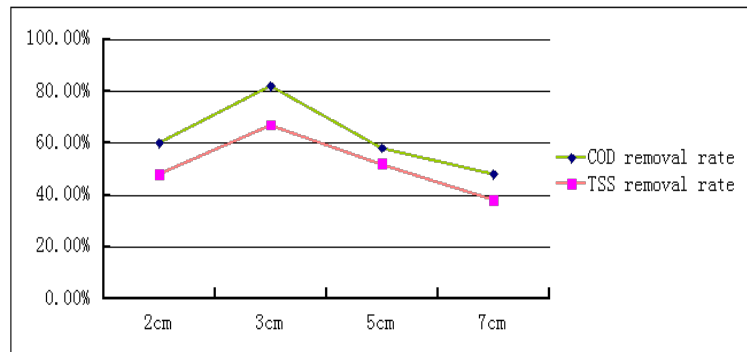


Fig. 3-4 The impact on the treatment effect of plate spacing

#### 4. Conclusion

1. Electric coagulation process of landfill leachate both for removal of COD and TSS have better results. And the device is simple and small footprint, material is convenient and available, operation time is short, ideally suited for oilfield produced water treatment of remote areas.
2. Using Fe as the anode and C as the cathode, adjust the water sample the initial PH value of 6, current density of  $9\text{mA}/\text{cm}^2$ , plate spacing of 3cm, electrolysis 1 hour, treatment effect is relative to the best. Sewage meets Secondary emission standards.

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