

The preparation of TiB₂ powder by carbothermal reaction using a DC electric arc furnace

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Abstract. The method of carbothermal reaction is used to prepare titanium boride (TiB₂) in a DC electric arc furnace by using a powder mixture of TiO₂, H₂BO₃ and C. The initial product is milled and the impurities are removed by pickling and reverse flotation. The final TiB₂ powder with a purity of over 99% can be obtained. The structure and impurities of the products are studied by X-Ray Diffraction, field emission scanning electron microscope (FESEM), energy dispersive spectrometer (EDS) and inductively coupled plasma (ICP).

Introduction

As a newly-developed inorganic nonmetallic material, titanium boride draws more and more attention of the researchers, because of its various advantageous properties and the widespread uses. Due to its high melting point (3225°C)[1], good conductivity[2] and the good wettability to melted aluminum, titanium boride could be processed on the anode in the aluminum electrolytic cell. Because of the high hardness (30 GPa)[3] and the low density (4.5 g/cm³), one of usage of TiB₂ is hard tools material. Titanium boride could be also used to in the manufacture of evaporation boats and PTC materials, because of its good conductivity and the good thermal shock resistance[1, 4]. At present, there are also restrictive factors in its wider applications, such as high price, high producing costs, low producing efficiency.

At present, the industrial methods of titanium boride preparation mainly includes self-propagating high-temperature synthesis, chemical vapor deposition method, and carbothermal reduction method by carbon tube furnace, etc[5]. Self-propagating high-temperature synthesis (SHS) has the advantages of lower reaction temperature, less energy consuming and smaller particle size, but leads to stable oxide impurities, which is difficult to be separated, moreover, maximization and automation of the equipment are also problems[6,7]. The chemical vapor deposition (CVD) is more appropriate to produce a small amount of TiB₂ production or film coating[8,9]. The carbon tube furnace carbothermal reduction method contains a problem of low efficiency, low energy utilization and equipment maximization, because a high temperature of more than 1500°C is needed in the carbothermal reaction[10].

The arc furnace carbothermal reduction method made some progresses in recent years, but there are still some remaining problems, such as the appearance of TiC impurities which is hard to be separated, the existence of C impurities which is hard to isolated[11]. In this paper, a series of process of TiB₂ powder preparation was researched including the arc furnace carbothermal reduction process, milling process and purification process. Because of the advantages like high and centralized temperature supplying that the arc plasma supplies[12], less TiC emerges. After the processes of milling, reverse flotation and pickling, TiB₂ powder with a purity of 99% and particle size of 5μm is obtained.

Experimental

The raw materials and their contents are given in the Table 1, and petroleum coke is used as the carbon source in this experiment:

Table 1 Raw material of arc furnace experiment

raw material	purity/%	Si	Fe	water	ash
H ₃ BO ₃	>99.00	<0.02	<0.01	-	-
TiO ₂	>98.00	<0.02	<0.01	<1.50	-
petroleum coke	-	-	-	-	<0.40

According to the thermodynamic calculation and the previous experiments, the ratio of H₃BO₃ should have a 30% overdose, and the ration of C should have a overdose of 10%, so the mass ratio of every raw material is TiO₂: H₃BO₃:C=1:2.015:0.825.

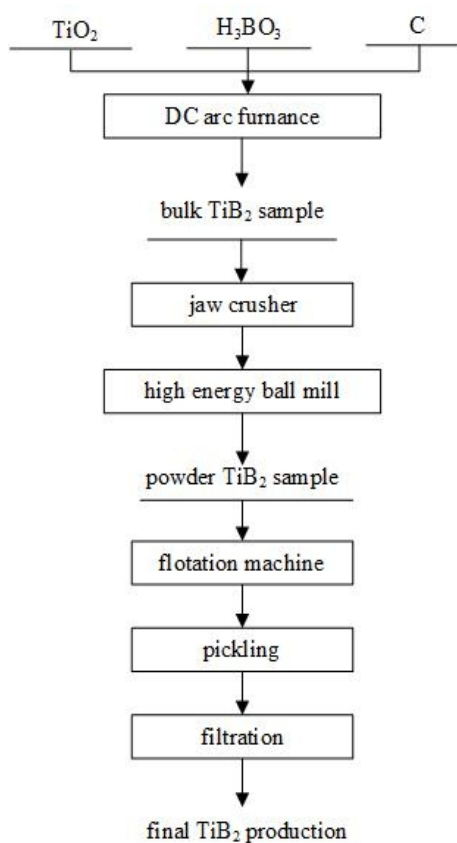


Figure 1 The process chart of TiB₂ powder preparation

The processes of experiment are as follows: 1. Mix the raw materials according to the ratio, and put them into the arc furnace to smelt; 2. When the smelting process is done, pull the graphite electrode out, and wait the bulk sample cooling together with the furnace; 3. Crush the bulk sample into little pieces with the jaw crusher, and then mill them into powders in an average size of 5 μm; 4. Use reverse flotation method and pickling method to purify the powder production, and finally attain a production with a purity of 99% and a particle size of 5 μm. The process chart is shown in the Figure 1.

Results and discussion

The results of smelting experiment. Figure 2 is the XRD pattern of bulk TiB₂ sample after the smelting process. As it is shown in the figure, in the method of arc furnace smelting, TiB₂ is prepared, which has an inclusion of C. In the mixing process, an extra 10% in the raw material powders is mixed, which aims to prevent the appearance of TiC. The pattern shows that there are no TiC peaks. The reaction happens in a very rapid way, due to the extremely high temperature of over 2000°C which is applied by the arc furnace [11]. According to the theoretical calculations, if enough or overdosed TiO₂ and C are applied, the reaction balance leads to the appearance of TiB₂. So the

overdose of boric acid and petroleum coke helps to prevent the appearance of TiC, and that is proved in the Figure 2.

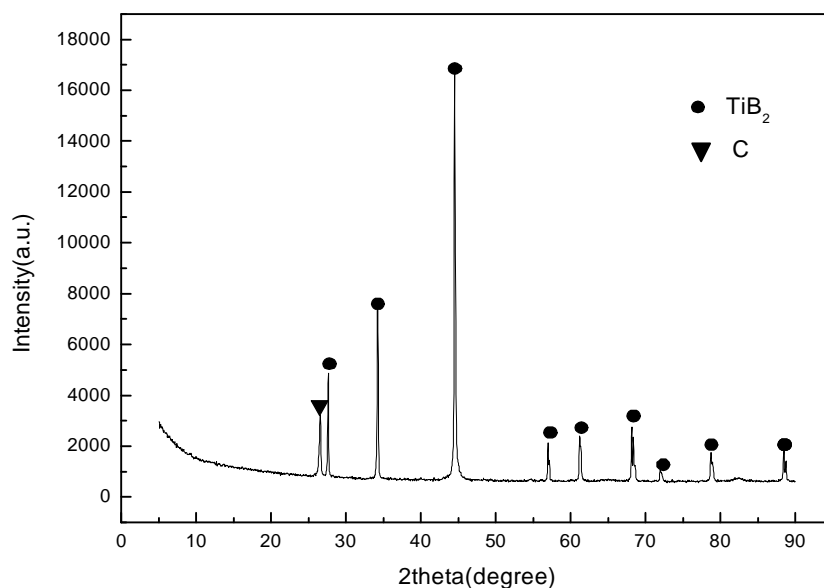


Figure 2 XRD pattern of TiB_2 sample after the carbothermal reaction

The SEM micrographs and EDS result of the smelting-prepared bulk TiB_2 sample are shown in the Figure 3(a) and (b). And as shown in the Figure 3(b), the EDS result of spot 1 proves that the impurity content is carbon. The Figure 3(a) displays that the bulk TiB_2 sample includes an impurity of C inside, which shows a morphology of irregularity and a scale of $10\mu m$. The impurity C is included in the uniform TiB_2 grains, and the grain boundaries between them are clear. Because of the extremely high temperature the arc plasma implies [11], the TiB_2 melts after the carbothermal reaction completes. Moreover, all these processes happen within seconds, which results in the uniform of TiB_2 grains and the singleness of the impurity grains.

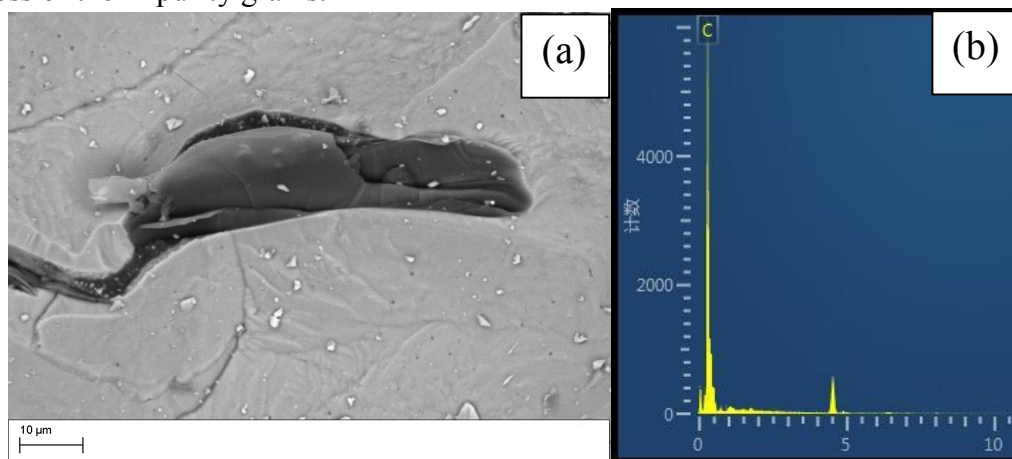


Figure 3 SEM micrograph and the EDS results of TiB_2 bulk sample

The results of reverse flotation. The carbon removal rates and recovery yields during the reverse flotation process are displayed in the Figure 4, in which the percentages of carbon are measured by the ICP method. The original percentage of C impurity is 5.67%, which decreases to 0.86% at the third time and 0.81% at the fourth time using the method of reverse flotation. The carbon removal rate increased greatly in the first reverse flotation process, in which the recovery yield is 76.9%. The carbon removal rate increases steadily but slowly in the 2 to 4 levels of reverse flotation process, however, the recovery yield keeps decreasing. As shown in the chart, the recovery yield decreases faster from the third time to the fourth (from 65.63% to 57.12%), however, it does not lead to an increase of carbon removal rate obviously. So 3 series of reverse flotation is the optimal selection in the practical operation.

XRD patterns of TiB_2 powder shown in the Figure 5 displays that the peak of C vanishes after reverse flotation, which means the reverse flotation method is a proper way of purifying TiB_2 prepared by arc furnace carbothermal reaction.

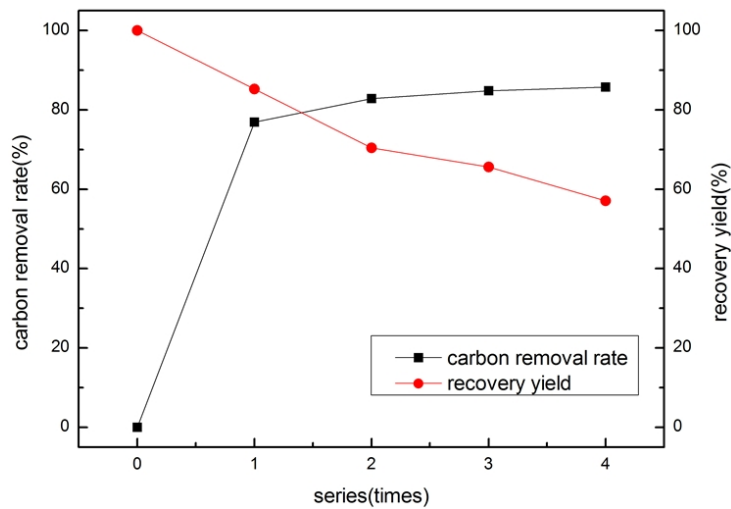


Figure 4 Carbon removal rate and recovery yield of TiB_2 samples at varies series of reverse flotation (a concentration of 30%, 0.075mol/L pine oil, pH=7)

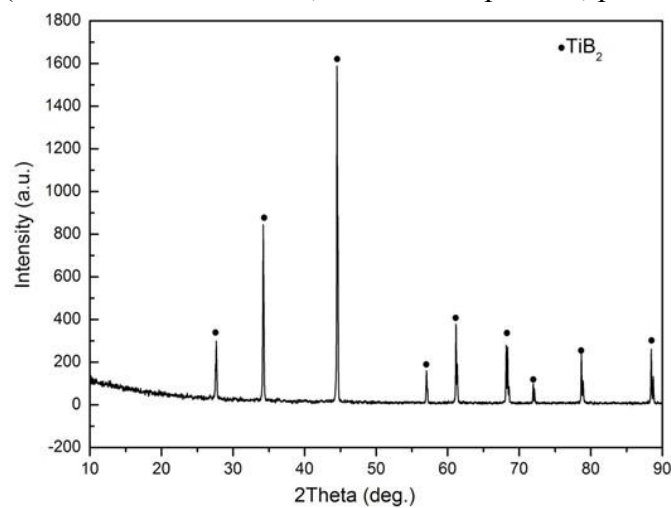


Figure 5 XRD patterns of TiB_2 sample after reverse flotation process

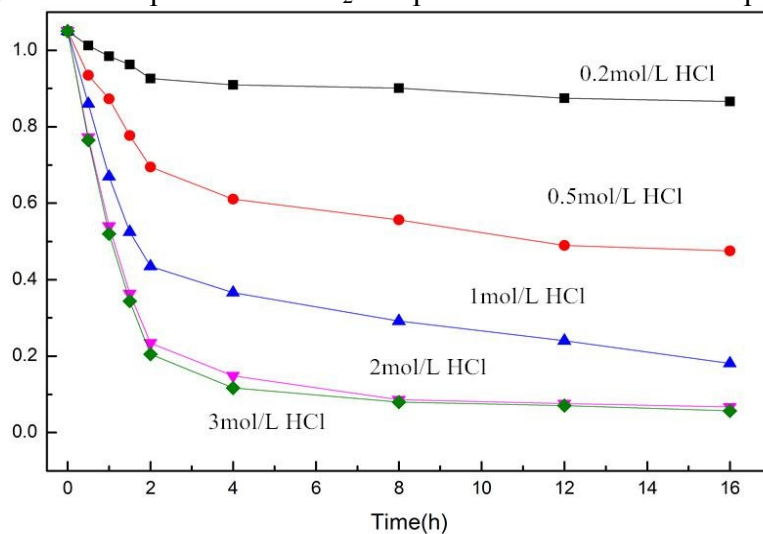


Figure 6 Concentration of Fe in different pickling conditions (various concentration of HCl and pickling time)

The results of pickling. As shown in Figure 6, the percentage of Fe impurity, which emerged during

the milling process, decreases from 1.05% to 0.057% at least using the method of HCl pickling. The decreasing tendencies varies when the concentrations of HCl and pickling time change, and the decreasing speed in the first 2 hours goes much faster, than later. 3mol/L HCl leads to the most obvious decrease to the minimum of 0.057% in 16 hours, in which the decreasing rate 94.57% attains. 2mol/L has a decreasing tendency barely the same as 3mol/L, reaching a concentration of 0.067% in 16 hours' pickling.

Conclusion

TiB₂ powder with a purity of over 99% can be prepared by the method of arc furnace carbothermal reaction. The TiC impurity is prevented by using a fixed raw material ratio, particle size and the mixing method. The C impurity is reduced from 5.67% to 0.86% after fourth reverse flotation. And then, the Fe impurity decreases from 1.05% to 0.067% by using 2 mol/L hydrochloric acid pickling for 16 hours. The results of ICP and XRD shows that TiB₂ powder with a purity of over 99% is obtained.

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