

## Effects of TiO<sub>2</sub> Contents on HA/TiO<sub>2</sub> Composite Coating by Electrophoretic Deposition

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**Abstract.** The HA composite coating with different TiO<sub>2</sub> contents was prepared by electrophoretic deposition in this study. The composite coatings were characterized by scanning electron microscopy (SEM), Ca/P atomic ratio in composite coatings was analyzed by Energy Dispersive Spectrometer (EDS), and the adhesion strength was analyzed by Universal Material Testing Machine. The results show that the density of HA-TiO<sub>2</sub> composite coatings was elevated with increasing TiO<sub>2</sub> contents after sintering at 850°C. The atomic ratio of Ca/P (calcium and phosphorus) of the HA/TiO<sub>2</sub> composite coatings showed a rising trend with the increasing contents of TiO<sub>2</sub> in suspensions. The atomic ratio of Ca/P was 1.69 and 1.75 which was close to the human bone (1.67), when the TiO<sub>2</sub> content was 10g/L and 15g/L in the suspension, respectively. The thickness of the coatings increased with the TiO<sub>2</sub> contents increasing, which was 30μm that deposited in the suspension content with 10g/L TiO<sub>2</sub>, and great combination was shown with the good uniformity of the coating. The adhesion strength was 9.10MPa as the TiO<sub>2</sub> content was 10g/L.

### Introduction

The hydroxyapatite (Ca<sub>10</sub>(PO<sub>4</sub>)<sub>6</sub>(OH)<sub>2</sub>, HA), is an inorganic constituent of the human bones, which can induce the growth of new bone<sup>[1-2]</sup>. It is a biomedical ceramic material that was widely used in material substitute to human hard tissues to improve the cell responses and osteo-conductivity, which can easily form a bond with the neighboring bone after planted in the human body; meanwhile, it is non-toxic, high bioactivity and high biocompatibility<sup>[3-4]</sup>.

Electrophoretic deposition (EPD) is an economical technique for the preparation of uniform HA coatings on titanium alloy substrate. The EPD technique has several advantages compared to other common methods of surface coating, including short formation time, simple apparatus and the shape of substrate is unconstrained etc.<sup>[5]</sup>. However, some shortage of the EPD method limited its application, such as the HA coatings is easy to thermal decompose into tricalcium and tetracalcium phosphates when the sintering after deposition at a high temperature, the sintering temperature should be set below 1000°C to retard the decomposition of HA coating<sup>[6]</sup>. What's more, the adhesion strength between HA and substrate is not high enough because of the mismatch of thermal expansion coefficient (CPE) between titanium ( $\alpha_{Ti}=8.7\times 10^{-6}K^{-1}$ ) substrate and HA coatings ( $\alpha_{HA}=(14\sim 16)\times 10^{-6}K^{-1}$ )<sup>[7]</sup>. The results showed that the transition of the HA coating that prepared between the substrate and the HA coating was with a proper thermal coefficient of expansion and can significantly improve the adhesion strength<sup>[8]</sup>.

In this paper, the TiO<sub>2</sub> was selected as transition layer because of the thermal expansion coefficient of TiO<sub>2</sub> ( $9.0\times 10^{-6}K^{-1}$ ) between titanium alloy and HA, and it can ease the residual stress. It will improve the performance of the combination between coating and the base. The functional HA-TiO<sub>2</sub> composite coating was fabricated by electrophoretic deposition and heat-treatment in the titanium substrate surface in order to study the effects of TiO<sub>2</sub> content on HA-TiO<sub>2</sub> composite coating performance.

## Experiment

### Experimental materials

Hydroxyapatite powder (30nm, Sinopharm Chemical Reagent, China) and TiO<sub>2</sub> powder (30nm, Sinopharm Chemical Reagent, China) were used as the raw materials. Ti-7.5Mo alloys (φ10×15mm) that prepared by gel-casting were used as metallic substrates in this study.

### Preparation of the Suspension and Deposition

N-butyl alcohol (Sinopharm Chemical Reagent, China) was used as a solvent and the pH of the suspension was adjusted to 5.5~6.5 by adding triethanolamine. In this study, four suspensions were prepared with different TiO<sub>2</sub> contents, shown as Table1. The content of HA was kept at 10g/L (1g HA was added in 100ml N-butyl alcohol) in each suspension.

Table1. Preparation of the suspension for HA-TiO<sub>2</sub> graded composite coatings

Contents of TiO <sub>2</sub> [g/L]	The amount of TiO <sub>2</sub> in the suspensions[g] (added in 100ml N-butyl alcohol)			
	Suspension a	Suspension b	Suspension c	Suspension d
5	0.5	0.3	0.1	0
10	1	0.6	0.2	0
15	1.5	0.9	0.3	0
20	2	1.2	0.4	0

### Characterization Testing

The microstructure of the coating was characterized by SEM, and the thickness of the coating was measured by observing the cross section morphology of coating. Elemental analysis of coating was evaluated by EDS, and the coating's atomic ratio of Ca and P can also be calculated.

Schematics of adhesion strength of the coating were showed as Fig. 1. The coating and the titanium substrate was bonded with the stainless steels respectively. Tension was applied on stainless steels till the coating was peeled from the substrate. The adhesion strength was figured by the following formula.

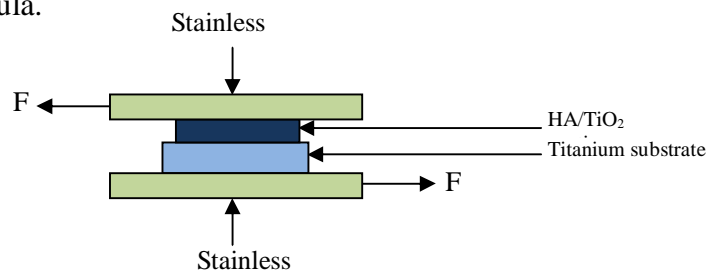


Fig.1. Schematic diagram of the adhesion strength test for the coating

$$\sigma = F/A \quad (1)$$

F- Maximum tension while coating peeled

A- Contact area of the stainless steel with the coating

## Results and Discussion

### Surface morphology and the Ca/P ratio of the HA/TiO<sub>2</sub> composite coatings

Fig.2 showed the surface morphology of the composite HA/TiO<sub>2</sub> coatings with different TiO<sub>2</sub> contents after sintering at 850°C. As can be seen, the coatings became non-uniform when the TiO<sub>2</sub> content increased to 15g/L and 20g/L. In fact, with increasing of the TiO<sub>2</sub> contents, the pore size and the number of the pore on the coatings reduced, the coatings became more compact, because a closely packed accumulation of HA and TiO<sub>2</sub> particles could be formed during a short time and the bonding between the HA and TiO<sub>2</sub> particles were strengthened by increasing the TiO<sub>2</sub> contents. The coatings could be porous and loose when the TiO<sub>2</sub> content was kept at a low level. However, the coating surface became non-uniform, which may be caused by the high potential of TiO<sub>2</sub>, the TiO<sub>2</sub> particles could be deposited more easy than the HA particles.

As can be seen in Fig. 3, there were little impurity elements on the surfaces of coatings. The atomic rate of calcium and phosphorus was 1.46 when the  $\text{TiO}_2$  content was 5g/L in the suspension, which was less than the human bone ( $\sim 1.67$ )<sup>[9]</sup> and illustrated that the crystalline of HA was not enough. When the  $\text{TiO}_2$  content was 10g/L and 15g/L, the Ca/P atomic rate was 1.69 and 1.75 respectively, which was close to the human bone and what can be seen that the appropriate  $\text{TiO}_2$  content in the suspension could promote the ossification of HA. However, when the  $\text{TiO}_2$  content was 20g/L, the Ca/P atomic rate was up to 1.92.

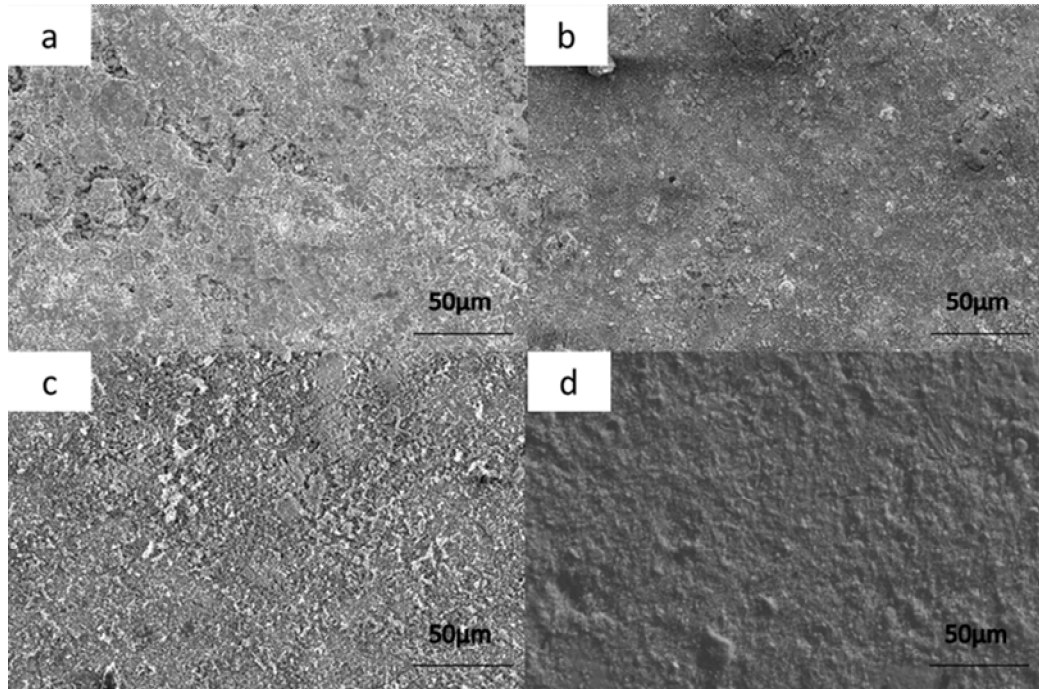


Fig.2. SEM images of the coatings deposited from the suspensions with 10 g/L HA and different  $\text{TiO}_2$  content (500×): (a 5g/L, b10g/L, c 15g/L, d 20g/L) after sintering at 850°C, respectively

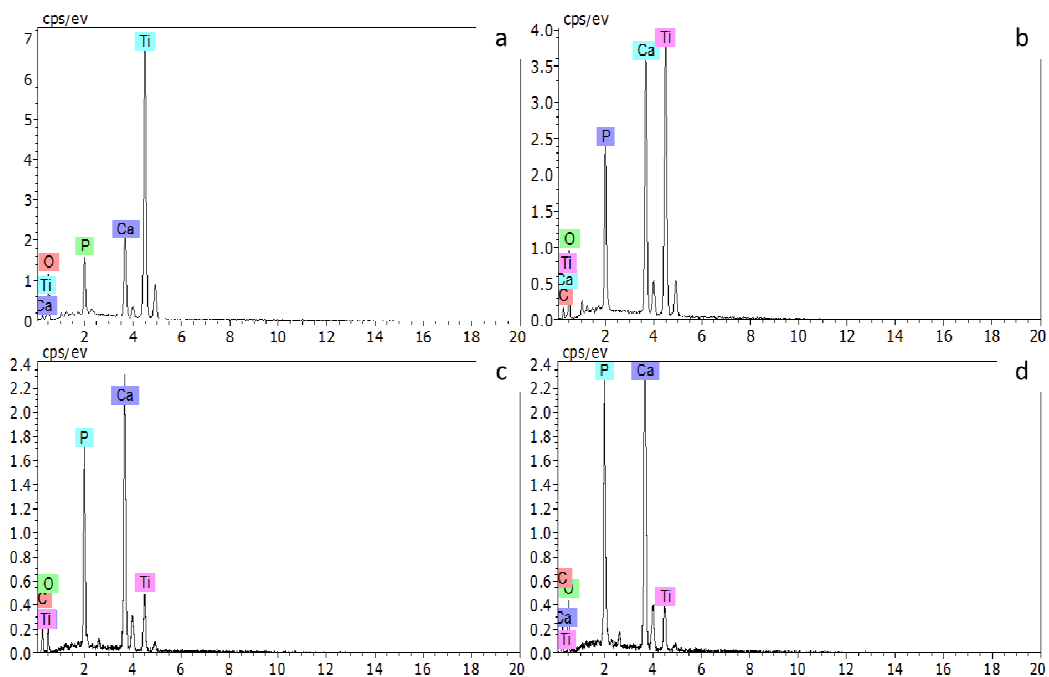


Fig.3. EDS of HA/ $\text{TiO}_2$  coatings with different  $\text{TiO}_2$  contents (a 5g/L, b10g/L, c 15g/L, d 20g/L) after sintering at 850°C

### Cross Section SEM micrograph of HA/TiO<sub>2</sub> coating

Fig. 4 was the cross section SEM micrograph of HA/TiO<sub>2</sub> coatings. It can be seen that the thickness of in the coatings increased as the TiO<sub>2</sub> contents increased. What can be known that the coatings were very uneven when the TiO<sub>2</sub> contents were 15g/L and 20g/L. What's more, the cracks were obvious, which could cause reducing adhesion strength greatly (Fig.4c and d). However, as can be seen thickness of coating was 25μm that deposited in the suspension content with 10g/L TiO<sub>2</sub>, and great combination was shown with the good uniformity of the coating.(Fig.4b).

The adhesion strength of the coatings with different TiO<sub>2</sub> contents sintering at 850°C were shown in Fig. 5. The adhesion strength of coating was merely 4.34MPa without any TiO<sub>2</sub>, and the adhesion strength was increased from 6.85MPa to 9.10MPa while the TiO<sub>2</sub> content was increased from 5g/L to 10g/L. However, the adhesion strength was decreased when the TiO<sub>2</sub> content was 15g/L and 20g/L, which was 6.18MPa and 4.08MPa respectively, which may be caused by the obvious cracks on the coatings.

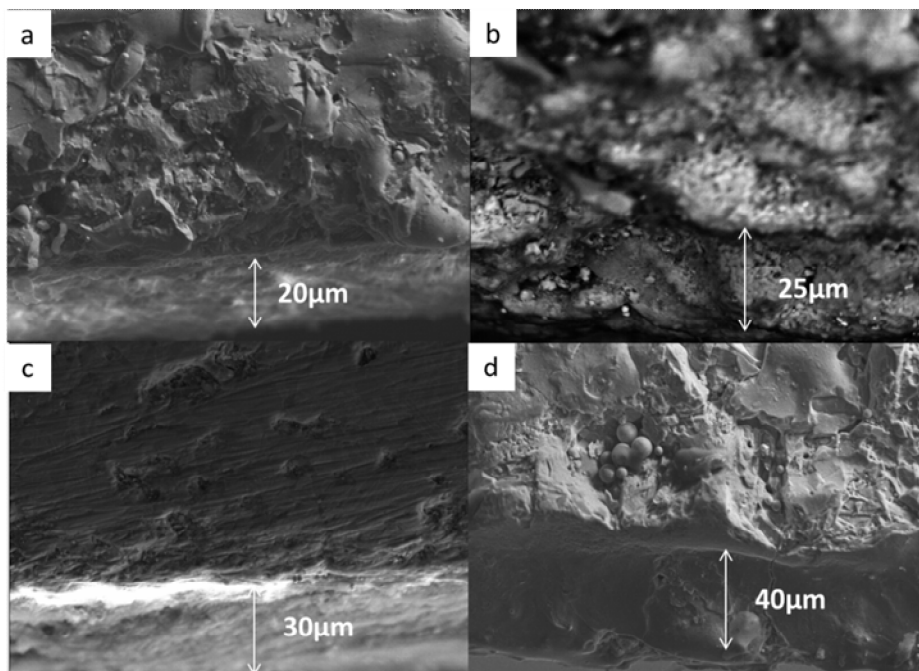


Fig.4. Cross section SEM micrograph of HA/TiO<sub>2</sub> coatings with different TiO<sub>2</sub> contents (a 5g/L, b 10g/L, c 15g/L, d 20g/L) after sintering at 850°C

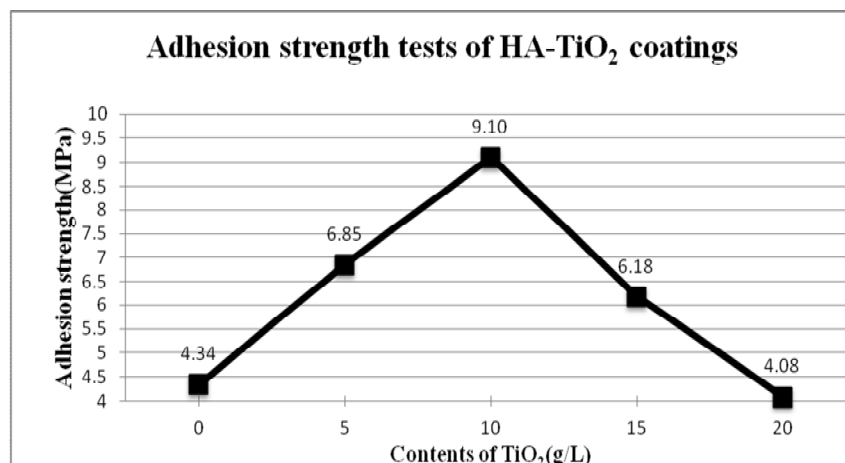


Fig.5. Adhesion strength tests of HA/TiO<sub>2</sub> coatings

## Conclusion

The different TiO<sub>2</sub> contents composite coating was successfully prepared by electrophoretic deposition in this paper. The Ca/P atomic ratio of the HA/TiO<sub>2</sub> composite coatings showed a rising trend with the TiO<sub>2</sub> contents in suspensions increased. When the TiO<sub>2</sub> content in the suspension was 10g/L and 15g/L, the Ca/P atomic rate was 1.69 and 1.75 respectively, which was close to the human bone. The thickness of the coatings increased as the TiO<sub>2</sub> contents increasing. The thickness of coating was about 20μm that deposited in the suspension with the TiO<sub>2</sub> content of 10g/L, and great combination was shown with the good uniformity of the coating. The adhesion strength was 9.10MPa while the TiO<sub>2</sub> content was 10g/L.

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