

Investigation of Ni-based SiC Thicker Coating by Induction Cladding

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Abstract— Fabrication of thicker coating by induction cladding could achieve excellent properties, such as wear resistance, corrosion resistance, heat resistance and so on, and also was an important technology for machinery parts remanufacturing. Used the mixture of Ni60 and SiC powder as coating materials, discussed the processing feasibility of preparing SiC coating on steel substrate by induction cladding. The optimum technological parameters by which can obtain good cladding quality were: SiC ratio is 20%, sodium silicate as the binder, heated at 150 °C for 1 hours, intermitted heating with Ar gas protection, and achieved high quality SiC coating: the layer thickness was above 400μm, coating hardness was about 750 HV, and achieved metallurgical bonding between the coating and steel substrate.

Keywords— induction cladding; SiC coating; coating capabilities; metallurgical bonding

I. INTRODUCTION

With the development of modern industrialization, the service performance of engineering equipments required more and more higher, and the machinery components, as the basic units of engineering equipments, were suffered more and more poor work environment. Preparing coating on the components to be repaired can achieve expected surface performances of abrasion resistance, corrosion resistance and other special properties, by reasonable design of the coating composition, and application of proper processing, and be benefit for improving material properties and prolonging the service life of all sorts of engineering components [1].

At present, research of functional coating materials focused on the cermets, which obtained ceramics properties, including excellent chemical stability, resistance to wear, corrosion, and high temperature, et al, meanwhile possessed the characteristic of the metal materials with high strength and toughness[2]. Nowadays, cermets coating materials, mainly including carbides, oxides, nitrides, borides and inter-metallic compounds, had been studied and applied in engineering[3]. There were two typical wear-resistant coating materials, one possessed the characteristic of high melting point and high

hardness, mainly included some carbides and nitrides, such as WC, B₄C, TiN etc [4]; Another was the cladding materials with low friction coefficient and self-lubrication, such as MoS₂ [5]; Borides which had capabilities of high melting-point, high hardness, chemical stability, etc, often used as protective coating materials to achieve special performances of good thermal stability, wear resistance and corrosion resistance [6]; Pure oxides such as ZrO₂, Al₂O₃, SiO₂, etc., or their composite were constantly used to prepare thermal barrier coatings. Among these, ZrO₂ was most widely applied[7]. Meanwhile, the coating fabrication technology were rapidly developed, such as thermal spraying and spray-welding[8], laser cladding[9], sol-gel process[10], Vapor deposition[11], etc. But most of the processing equipments were more expensive, the cladding layers were usually thin, and the junction between the coating and substrate mostly showed the physical binding mode, resulted in the lower bonding strength. Therefore, all above technologies were not suitable for remanufacturing of engineering components, which needed enough coating thickness to recover its dimension or shape defects.

II. MATERIALS AND EXPERIMENTAL METHODS

Selected SiC powder as coated aggregates. SiC powder possessed excellent properties, such as density of 3.20~3.25g/cm³, melting point of about 2700°C, micro hardness of 2840~3320kg/mm², and had stable chemical properties, high thermal conductivity, low thermal expansion coefficient, good wear resistance and so on. Thus, using SiC powder for cladding could greatly improve the wear and corrosion resistance and other properties [8]. In order to make the SiC powder effectively consolidate on the surface of 45 steel substrate, Ni60 was used as a transition material, which improved to make the metallurgical reaction between the Ni60 alloy and the steel substrate [9].

Experimental procedures as follows: (1) surface pre-treatment of steel substrate, including the elimination of surface rust, oil and other impurities, surface activation; (2) mixed the Ni60 and SiC powder materials to be well-distributed, and added proper proportion of the proportion of

blindings to form the Viscous paste, then coated on the steel substrate; (3)Dried the coated specimens; (4) Placed the specimens in induction device, and heated to the molten state, stop heating; (5) took out the specimen and air-cooled to room temperature; (6) detected the coating performances.

Using high frequency induction equipment of TG-GP-100KW models to carry out the coating forming process, MiniTest740 coating thickness gauge to measure the coating thickness, optical microscopy or scanning electron microscopy (SEM) to analyze the microstructure, scanning electron microscope (SEM) to scan the interfacial chemical composition distribution, and TWV-1S type micro-accelerometers to measure the surface hardness and interfacial hardness distribution.

III. RESULTS AND DISCUSSION

A. Effect of SiC mass ratio

Fig.1, 2 and 3 showed the bonding characteristics of the different SiC mass ratios which being respectively 10%, 20% and 30%. The results showed that the bonding transition layer was all formed between the substrate and coating layer, and all achieved metallurgic bonding effects. Ni60 Alloy contains many metallic elements, such as Ni, Cr, Fe, etc., which can solid dissolved in steel matrix, so these elements could diffuse into each other between Ni60 Alloy and 45 steel substrate when heated to higher temperature and kept a certain time, so created the metallurgical bonding, and obtained good bonding strength. But with the increase of the SiC mass ratio, the thickness of transition layer regions decreased. The principal reason was that, SiC powder had good chemical stability, and its numerous existence in Ni60 Alloy could prevent the diffusion of Ni, Cr, Fe, etc.. Fig.4 showed the result that, the surface hardness increased with the increase of SiC powder mass ratio, and indicated that produced the particle-reinforced effect in the Ni60 alloy due to the existence of SiC particles with high hardness. Comprehensively considered the combination state and the coating performance, the best coating quality could be achieved when the SiC mass ratio was 20% in coating mixture.

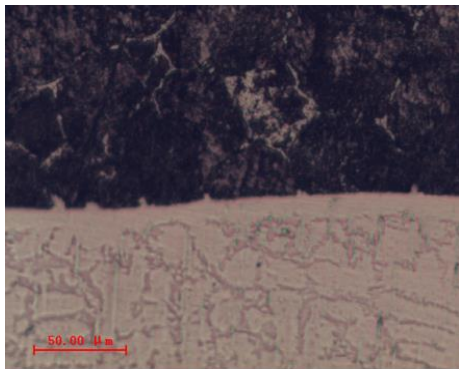


Fig. 1 bonding pattern of 10% SiC mass ratio

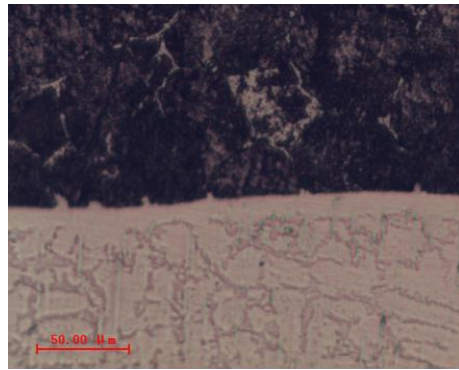


Fig. 2 bonding pattern of 20% SiC mass ratio

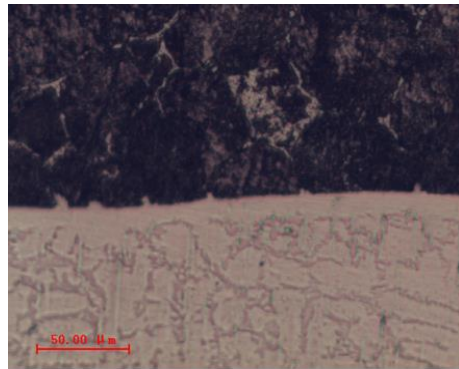


Fig. 3 bonding pattern of 30%SiC mass ratio

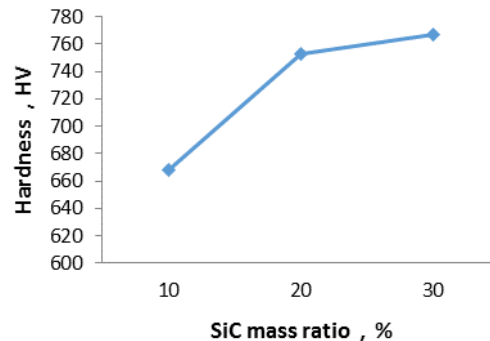


Fig. 4 surface hardness of different SiC mass ratio

B. Effect of binder types

Fig. 5 and Fig. 6 respectively illustrated the bonding morphology of the samples which was prepared when using turpentine (mixed rosin and Turpentine together in the mass ratio of 1:3) and sodium silicate as binder. The metallurgical bonding can be obtained by using sodium silicate as binder, as shown in Fig.5; whereas obvious dividing boundary occurred between the coating and substrate when using turpentine as binder, as shown in Fig.6. Turpentine was organic compound, and easy to produce large amounts of gas when being heated. Thus easily produced a certain air gap in the interface, and cut off the connection between coating and substrate. Accordingly, even though in a highly heated state, the diffusion of elements could not effectively carry out, and difficult to achieve the metallurgical bonding effect. In the experimental process,

found that the bulging was frequently happened, and some white smoke was also emerged in the course of induction heating. But using sodium silicate which was inorganic compound and with low moisture content, as binder seldom evaporated the moisture, thus could maintain the close contact between coating and substrate in the process of induction cladding, better for the elements diffusion of Ni, Fe, and easily achieved good metallurgical combination.

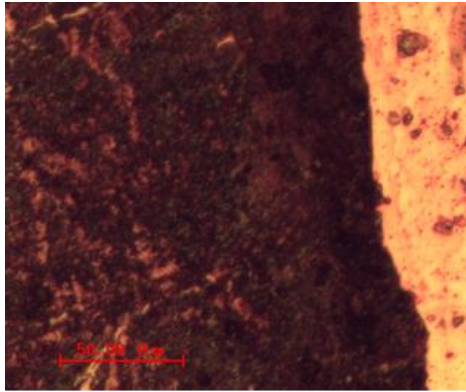


Fig. 5 bonding pattern of sodium silicate as binder

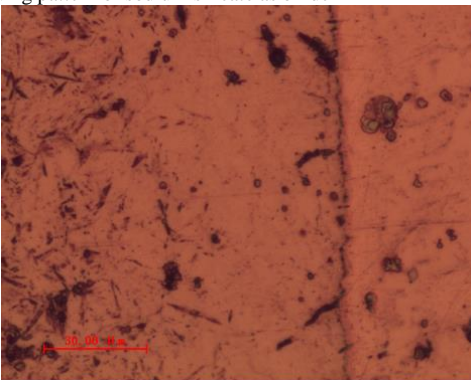


Fig. 6 bonding pattern of turpentine as binder

C. Effect of coating drying methods

The effect on the cladding quality of coating drying methods, including air dried and preheating at 150°C for 60 min, was also important. As illustrated in Fig.7, a distinct boundary existed in the joint region between the coating and substrate, proved that no obvious transition layer was produced in processing of air-dried at room temperature. For the air dried process, the drying procedure began from the outer surface, thus the moisture existed nearby the coating-matrix interface was difficult to escape out, easier to form a air gap in the subsequent induction heating, thus contributed to the lack of elements diffusion between the coating and the substrate. Whereas for the preheating-drying process, internal moisture existed in coating mixture could escape out in a steady flow, and the moisture emissions could be carried out completely. But if the temperature was above 200°C, surface bulging was obvious produced due to the excessive gas escaping current, even caused the coating materials falling off. Proved by experiments that, achieved excellent bonding patterns when preheated at 150°C for 60 min, and formed a clear transition region, as schematic in Fig. 8.

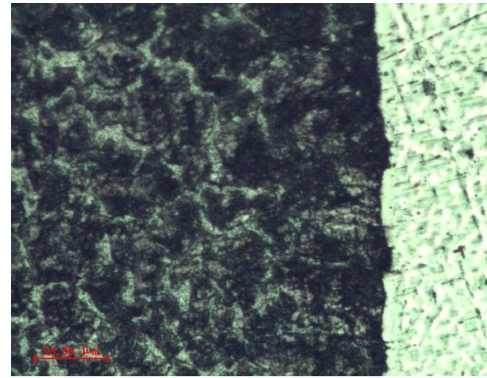


Fig.7 morphology of air dry for 24 hours

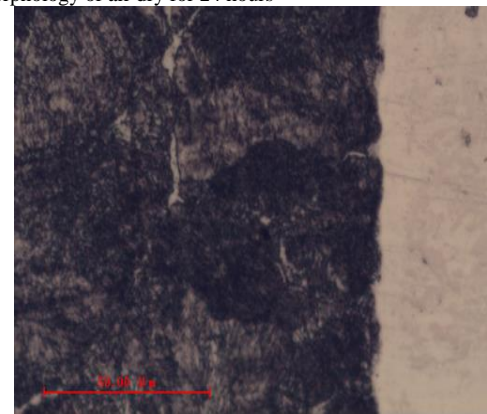


Fig. 8 morphology of 150°C preheating for 60 min

D. Effect of induction heated methods

The effect of the heating parameters in the induction cladding process on the coating quality was obvious. Fig.9 described the changing of the coating thickness obtained by three different heating procedures : (1) continuous heating in a constant current; (2) intermittent heating process which heated for 2s in a certain current, stopped heating for 5s or so, then reheating again, until occurred molted state on the surface; (3) intermittent heating with argon protection. For continuous heating process, because of the strong heating intensity, the internal residual moisture would be vaporized and expanded rapidly, caused the coating materials falling off from the sample surface, so the coating thickness is smaller. As for the intermittent heating, the intensity of heating were declined, the residual moisture could escape slowly; Meanwhile in the intermittent heating process, the relative high temperature state on the sample surface would stay for a longer time, the elements diffusion between the coating materials and substrate carried out more adequately, so the coating thickness was increased greatly; Adding the argon protection measures for the intermittent heating process avoided the oxidizing reaction of the coating material, and more effectively prevented the coating powder materials fall off, so the coating thickness still raised up; thus, formed excellent metallurgical bonding, and basically eliminated the oxidation inclusions within the transition layer, as shown in figure 10.

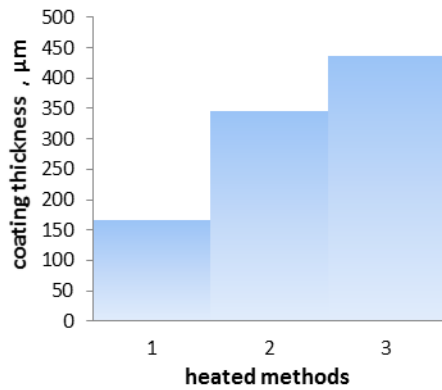


Fig. 9 comparison of coating thickness. 1-continuous heated; 2- intermittent heating; 3-intermittent heating with argon protection.

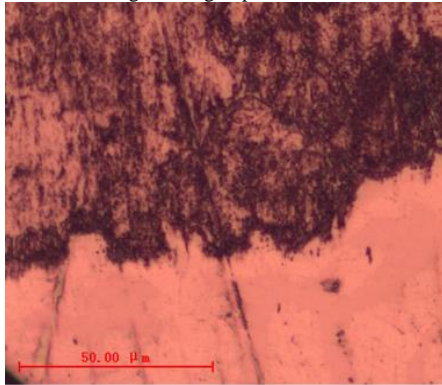


Fig. 10 morphology of intermittent heating with argon gas

E. Optimized process parameters and the cladding quality

Through large amounts of experiments, achieved the comparatively ideal processing parameters of getting desired cladding quality, including: (1) SiC mass ratio was 20% in composite coating material; (2) used sodium silicate as binder; (3) preheated at 150°C for 60 minutes to remove the internal moisture; (4) induction heated in intermittent heating style with the aid of argon protection.

1) Cladding process had little effect on the substrate microstructure

Fig.11 showed the original microstructure of the 45 steel substrate, and the morphology of the substrate after cladding process as schematic in Fig.12. Compared Fig.11 and Fig.12, the ferrite and pearlite grains basically remain unchanged after induction cladding process, and resulted little damage to the substrate. This is due to the very short time of induction heating, and the heating energy was concentrated on the surface thinner layer.

2) Displayed clear metallurgical bonding features

Fig.13 showed the microstructure of the junction area. Compact structure and fine grains existed in the cladding layer, and the distribution of SiC particles is uniform; produced a clear transition layer, which produced by elements diffusion of Fe, Ni and so on. The component scanning curves represented that, both Fe and Ni elements expressed the features of transitional changes in the

bonding area, as shown in Fig.14; The results of hardness testing also reflected the transitional features of mechanical properties, as shown in Fig.15. These results indicated that excellent metallurgical bonding was produced, and achieved formed closely integrated and fine microstructure.

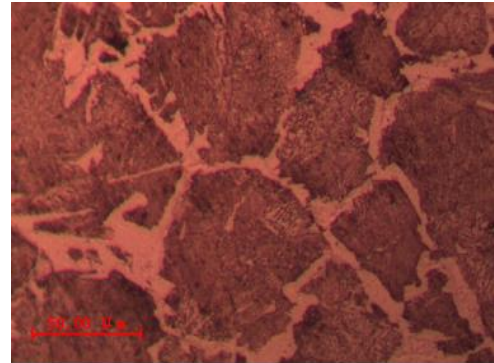


Fig. 11 microstructure of the primary substrate

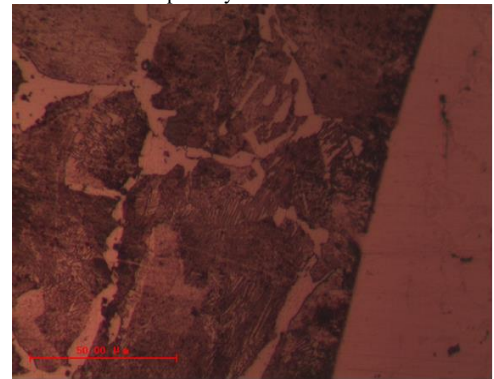


Fig. 12 microstructure of the substrate after coating

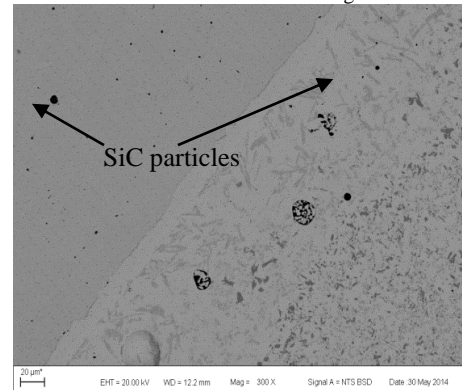


Fig. 13 microstructure of the transition layer

3) achieved the desired performances

The thickness of the cladding layer could reach over 400μm, and basically maintained technological requirements of engineering components repairing and remanufacturing process; The hardness in cladding layer could be up to 750 HV, and afford good wear resistance for the repaired material surface; Corrosion testing of 5% hydrochloric acid for 8 hours also proved that, the formation of SiC coating greatly improved the corrosion resistance of substrate materials, as shown in Fig.16.

IV. CONCLUSIONS

Carried out feasible research on the technology of Ni based SiC coating preparation by induction cladding process, and drawn the following conclusions:

(1) The best technological condition of Ni based SiC coating preparation was that: SiC mass ratio in the coating material was 20%; using sodium silicate as binder; well mixed Ni60 and SiC powder to be a paste fluid, then coated on the substrates; preheated at 150°C for 60 minutes; heated in intermittent induction heating style with the aid of argon protection, stopped the heating process until the surface melting signs appeared, air cooled to the room temperature.

(2) achieved excellent metallurgical bonding; the surface hardness reached HV750, the coating thickness was above 400μm; improved the properties of corrosion resistance.

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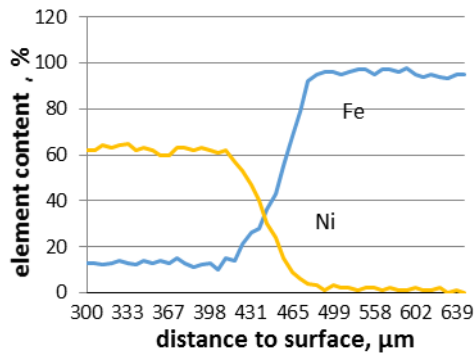


Fig. 14 component scanning curves of Fe and Ni elements

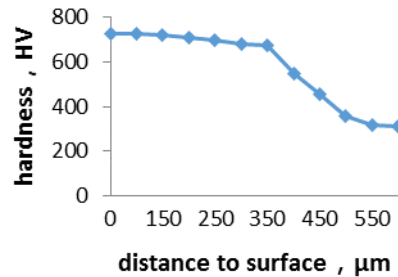


Fig. 15 curve of hardness variation along radius

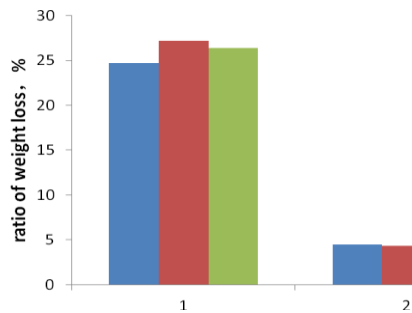


Fig. 16 comparison of weight loss in corrosion experiment. 1-Non-coating sample; 2- sample with coating