

Facile Preparation of Few-layer MoS₂-NS by Liquid-Phase Ultrasonic Exfoliation

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Abstract: In recent years, two-dimensional layered nanomaterials have attracted wide attention due to their unique physical, chemical, electronic and mechanical properties. Among them, ultrathin molybdenum disulfide nanosheets (MoS₂-NS) are one of the important two-dimensional nano-functional materials. In this paper, ultrafine MoS₂-NS were prepared by a simple liquid-phase ultrasonic exfoliation. Molybdenum disulfide powder was used as the raw material, sodium hydroxide and N-methyl-2-pyrrolidone were added as the penetrant. After vacuuming, few-layer MoS₂-NS were prepared by ultrasonic method at ultrasonic power of 360W for 6h. The microstructure were observed by TEM, the phase structures were characterized using XRD and Raman spectrometer.

Introduction

Two-dimensional nano material is believed to have atomic level or molecular level thickness and infinite plane size of a new class of materials. In 2004 Novoselov and Geim by micro-mechanical stripping from the graphite peel off the graphene, which is the most representative of the two-dimensional materials[1]. Graphene is constituted by carbon atoms in a single layer with a honeycomb structure[2]. The properties of Graphene are different from massive graphite such as mechanical properties, electrical properties, optical properties and thermal properties[3]. Scientists are interested in the two-dimensional materials for example transition metal disulfides which have a graphene-like structure, because the two-dimensional structural material has bright prospects[4-5].

The graphene-like molybdenum disulfide (MoS₂) is an important two dimensional layered material, molybdenum disulfide's crystal structure is hexagonal crystal system, but the single-layer MoS₂-NS is constituted by three atomic layers by covalent bonds, Molybdenum atoms sandwiched between the upper and lower layers of sulfur atoms, similar to a sandwich structure[6]. Few-layers MoS₂ is constituted by a few monolayer MoS₂ with a weak Vander Waals force. The distance between the layers is about 0.65 nm[7], So the interaction force is very strong, and the interaction between layer and layer is weak, this particular layered structure predetermines the excellent lubricating and catalytic properties of MoS₂-NS[8-9]. The band gap of MoS₂-NS is about 1.2eV, and the energy band is changed from indirect band gap to direct band gap with the number of layers decreases, and the direct band gap transition of the monolayer MoS₂ reaches a maximum value of about 1.9 eV. The graphene-like MoS₂ has a controlled band gap and has a brighter prospect in optoelectronic devices[10-11].

However, it is difficult to prepare the graphene-like MoS₂-NS with few layers by chemical and physical methods, the method of prepare molybdenum disulfide at home and abroad have the following categories: Micro mechanical stripping method, Lithium ion intercalation method, Liquid ultrasonic method, chemical vapor deposition and Hydrothermal method. Micro-mechanical stripping method which use special adhesive tape to stripping nanosheet is a simple method of

obtaining two-dimensional materials. In 1965 Frindt[12] used a special tape to peel off a few layers of MoS₂-NS, This method is very effective to prepare monolayer materials. The method is simple, rapid, and has a high stripping rate, and the stripped MoS₂-NS has a high electron mobility. However, the micro-mechanical stripping method has a very low production efficiency. In 1986, Joensen[13] firstly prepare monolayers of MoS₂-NS by lithium ion intercalation. This method is more complex, the reaction time is longer, and it also can changes the crystal structure of MoS₂-NS. YShi[14] prepared graphene-like MoS₂-NS on a graphene substrate by chemical vapor deposition (CVD). The advantage of this method is pure MoS₂-NS product, the disadvantage is that the reaction conditions have a great impact on the product. Coehoorn R[15] prepare monolayer MoS₂-NS by ultrasonic the N-methyl-2-pyrrolidone (NMP) and MoS₂ mixed solution. Although liquid phase ultrasonic stripping method is inefficient, the product had a complete structure, this method can be used for large-scale production.

In this paper, we sought a suitable reproducible experimental condition by a large number of experiments, and a simple liquid-phase ultrasonic exfoliation was used to prepare the ultra-thin molybdenum disulfide nano-sheets for other research in different fields .

Experiments

Preparation of Few-layer of MoS₂ Nanosheets:

The chemical reagent of experiment are analytically pure, the MoS₂ powder was purchased in Shanghai Aladdin Biochemical Technology Co.,Ltd. the NMP was purchased from Tianjin DaMao Chemical Reagent Factory. The Sodium hydroxide (NaOH) was purchased from Tianjin Tianli Chemical Reagent Co.,Ltd.

Weighted in different MoS₂/NMP/NaOH mass ratios, they were then dissolved in 40 ml of DI water, then the solution was vacuum-treated with a special vacuum pump, and the NMP penetrated sufficiently into the MoS₂ sheet and let it rest for more than 48 hour. The solution was transferred to ultrasonic cleaner, then treated the solution with different ultrasonic power and different ultrasonic times. Followed by centrifugal separation (3500 r/min), the supernatant was collected and characterized.

Characterization:

Morphology of the samples were observed by transmission electron microscope (TEM, Hitachi G-7650). The phase structure of the final products was characterized using X-ray diffraction (XRD, BRUKER-AXS, Germany). In addition, a Raman spectrometer (LabRAM HR Evolution) was used to obtain the corresponding Raman spectra of the few-layer MoS₂-NS.

Results and Discussion

It is known from the literature that the NaOH contributes to stripping of the graphene-like MoS₂-NS. The graphene-like MoS₂-NS sample were prepared at the content of NMP is constant, MoS₂/NaOH with mass ratios of (a) 1/0 (b) 2/1 (c) 1/1 (d) 1/2. Representative TEM images of the graphene-like MoS₂-NS sample are shown in Fig.1. As shown in Fig.1(a-c), With the increase of the amount of NaOH, the graphene-like MoS₂-NS gradually becomes thinner, but after a certain proportion, the graphene-like MoS₂-NS appears wrinkle, As shown in Fig.1(d).

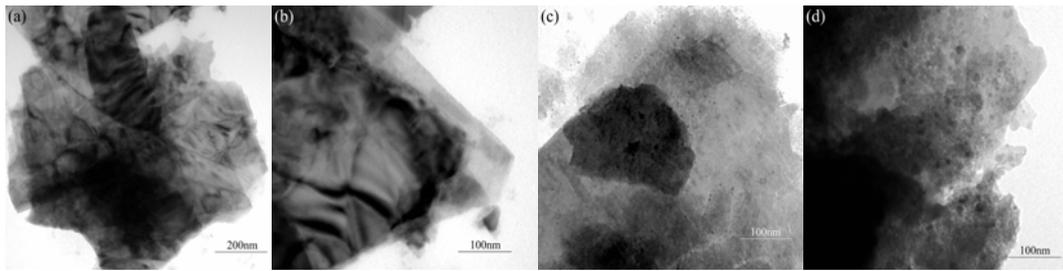


Figure 1. Representative TEM micrographs of MoS₂-NS with different mass ratios of MoS₂/NaOH:
(a) 1 / 0 ,(b) 2/1 ,(c) 1/1,(d)1/2 .

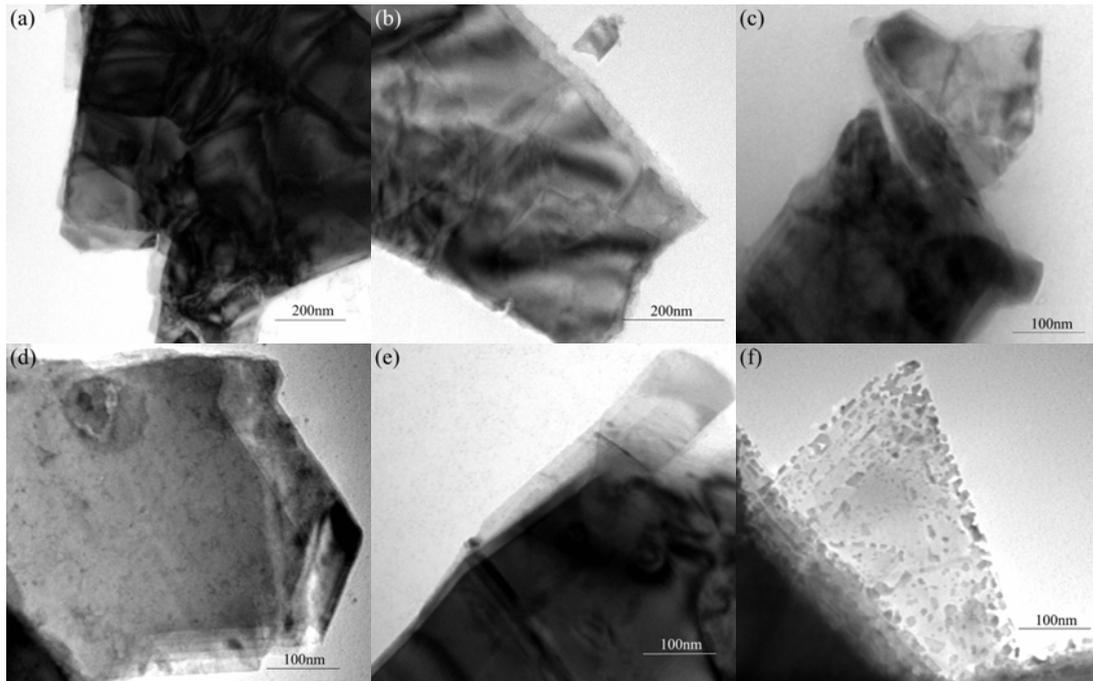


Figure 2. Representative TEM micrographs of MoS₂-NS with different Ultrasonic power
(a)180w,(b)240w,(c)300w,(d)360w,(e)480w,(f)600w.

The ultrasonic cleaner will cause the liquid to vibrate and produce tens of thousands of tiny bubbles. the collapse of the bubbles during the longitudinal propagation of the ultrasonic waves creates a tremendous pressure that will damage the surface of the material. In this experiment, the ultrasonic wave transferred to the NMP in MoS₂ layer, and made the NMP explode that a series of small explosions were also produced to impact the MoS₂ layer gradually to separate the MoS₂ layer from the MoS₂ block. As show in Fig.2,the ultrasonic power has a great influence on the peeling of MoS₂-NS.With the increase of the ultrasonic power, the number of the peeling layers decreases gradually, but when the ultrasonic power reaches a certain value, the stripping effect will be destroyed sample. As shown in Fig.2(a),when the ultrasonic power is 180w,the molybdenum disulfide samples prepared are relatively thick as a whole, there is no obvious change; As shown in Fig.2(b),when the ultrasonic power of 240w,the whole began to thin; As shown in Fig.2(c),when the ultrasonic power of 300w;As shown in Fig.2(d),when the ultrasonic power is 360w,the whole thinning; As shown in Fig.2(e),when the ultrasonic power of 480w only local thinner, the majority is still thick; As shown in Fig.2(f),When the ultrasonic power of 600w ultrasound, due to the power is too large, the thin part of the shattered, the sample is destroyed.

If the ultrasonic time is short, the effect of peeling can not be achieved; and if the ultrasonic time is too long, the stripped MoS₂-NS will be shattered. As shown in Fig.3 (a-d),the thin layer of graphene-like MoS₂-NS is gradually thinned with the increase of ultrasonic time, but the edge of

sample will be damaged when the ultrasonic time is too long.

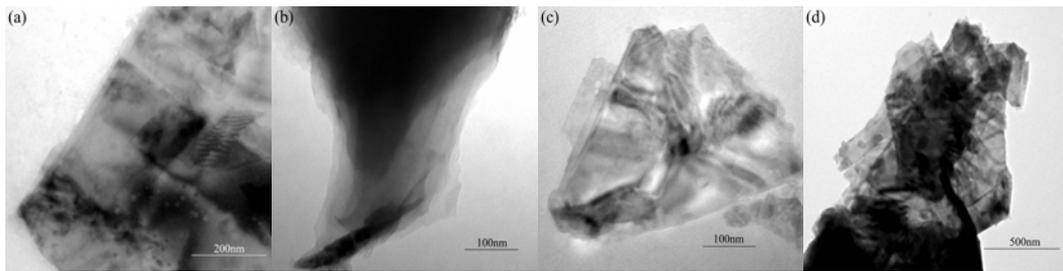


Figure 3. Representative TEM micrographs of MoS₂-NS with different Ultrasonic time

(a)360w 1h,(b)360w 2h,(c)360w 6h,(d)360w 10h.

After all the experiments were repeated to sum up the optimal parameters, a graphene-like MoS₂-NS sample was prepared by a simple liquid-phase ultrasonic exfoliation and characterized by transmission electron microscopy (TEM), as shown in Fig.4, using a simple liquid-phase ultrasonic exfoliation to produce a few or even a single layer of graphene-like MoS₂-NS. Through experiments we can see that we can see that graphene-like MoS₂-NS can be prepared by ultrasonic for 6h under ultrasonic power of 360W under the condition of controlling the mass ratio of MoS₂/NaOH was 1/1.

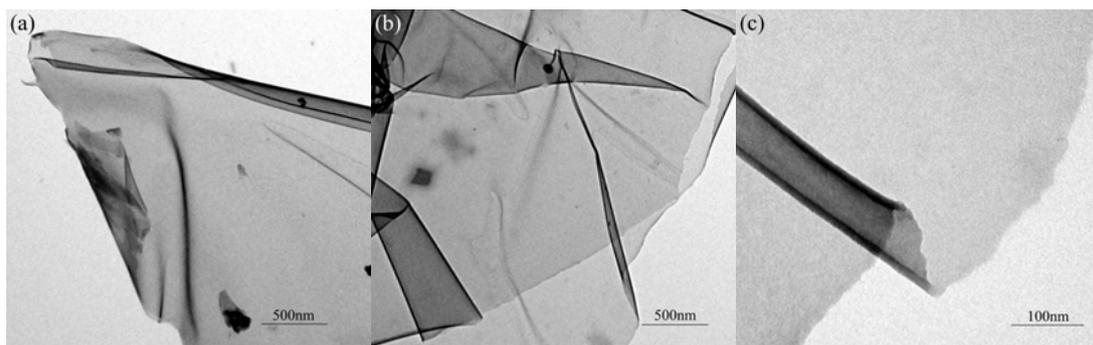


Figure 4. Representative TEM micrographs of MoS₂-NS with optimum conditions

Raman spectroscopy is the most effective tool to characterize lamellar compounds. The MoS₂ Raman spectra can be judged directly by measuring the wave number corresponding to Raman peaks. The MoS₂ Raman spectra have two basic vibration modes: E₁2g Mode and A₁g mode. E₁2g is the in-plane vibrational mode, which is the vibration of the two S atoms and the middle Mo atoms in the opposite direction. A₁g is the interlayer vibration mode, which is the vibration of S atoms and Mo atoms in the direction perpendicular to the layer. As the number of atomic layers decreases, the MoS₂ Raman vibrational mode E₁2g blue-shift, A₁g red shift, which can identify the MoS₂ layers. As shown in Fig.5(a), the number of layers of the graphene-like MoS₂-NS prepared by the simple liquid-phase ultrasonic exfoliation was decreased. From Fig.5(b), the peak position of MoS₂ is consistent with the peak position of MoS₂ powder. The diffraction peaks of MoS₂ powder correspond to 2H-MoS₂(JCPDS No.87-2416) The diffraction peak of (002) plane is lower than that of undissolved MoS₂, and the diffraction peak is obviously widened.

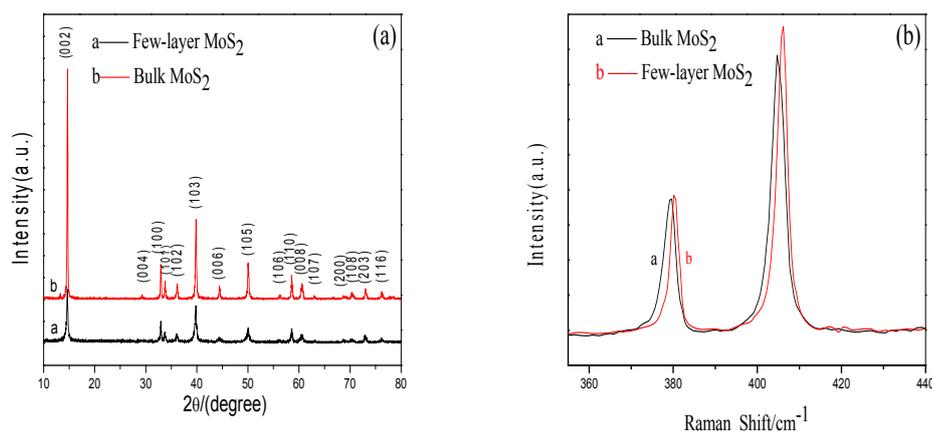


Figure 5. The XRD and Roman of MoS₂ with different scales

Conclusion

The results of TEM and Roman analysis show that the ultrasonic power, ultrasonic time and the amount of NaOH added directly affect the preparation of MoS₂-NS. With different test conditions, the degree of peeling of the prepared MoS₂-NS varies greatly. The optimal experimental parameters were obtained by comparing the experimental results. MoS₂ powders were placed in NMP and subjected to multiple pumping treatments. The ultrasonic power was 360w and the ultrasonic time was 6h.

Acknowledgments

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