

# Neural prediction model for microwave calcination-sulphuric acid leaching of germanium from zinc oxide dust

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**Abstract.** Based on the study of artificial neural network, the neural model was established for the prediction of germanium extraction from zinc oxide dust by microwave calcination-sulphuric acid leaching. Microwave heating temperature, liquid-solid ratio, leaching time, initial concentration of sulphuric acid and leaching temperature were the significant factors for the process. The results indicated that the neural network prediction model was reliable, the forecast and actual values fitted well. The model could be used to predict the regeneration experiments with high credibility and practical significance. The accuracy of convergence of the model has reached 10<sup>-5</sup>.

## Introduction

The neural network technology as a new technology of artificial intelligence is one of the frontier research direction of the rapid development of the international. The neural network is widely used in neurons are connected to each other into the complex network system, reflects a kind of human brain nervous system simplification, abstraction and simulation. With the development of research, neural networks have been widely used in telecommunications <sup>[1,2]</sup>, biomedical engineering <sup>[3,4]</sup>, chemical engineering <sup>[5,6,7]</sup>, automatic control[8,9,10], and expert systems<sup>[11,12,13]</sup>.

Neural network is a dynamic network system by a large number of neurons interconnected by highly nonlinear, adaptive, strong self-learning and self-organizing ability and massive parallelism and fault tolerance, so it has many traditional signal and information processing technology and not the advantage <sup>[14]</sup>.

In the present study, a neural model was established for the predicting the leaching ratios of Ge extraction from ZnO dust through microwave calcination-sulphuric acid leaching process.

## Establishment of neural network model

The effects of microwave roasting- sulphuric acid leaching process on germanium (Ge) extraction from zinc oxide smoke (ZnO) contain microwave heating temperature (A,  $^{\circ}$ C), liquid-solid ratio (B, mL/g), leaching time (C,h), initial concentration of sulphuric acid (D, mol/L) and leaching temperature (E, $^{\circ}$ C). So the input of the prediction model for the micro distribution of microwave heating temperature (A), liquid-solid ratio (B), leaching time (C), initial concentration of sulphuric acid (D) and leaching temperature (E), the range of control input, each test under the condition of germanium leaching ration (Y, %) as the output of the prediction model. Table 1 showed the input and output data.



N/I10POW0V0		
liquid-solid concentration	Leaching	Leaching
Run heating ratio (B Leaching of sulphuric	temperature	ratios
temperature $\frac{\operatorname{rand}(D, \operatorname{time}(C, h))}{\operatorname{mL}(q)}$ time (C, h) acid (D)	(F °C)	(V %)
(A, °C) $(A, °C)$ $(A, °C)$ $(A, °C)$ $(A, °C)$	(L, C)	(1,70)
	0.0	<b>50.0</b> (
1   310   8   4   12.6	80	50.86
2 290 6 3 9.5	70	52.75
3  290  6  3  9.5	/0	30.3 55.39
4 $2/0$ $8$ $4$ $12.05 210 4 2 12.6$	80	55.28 56.16
5    510    4    2    12.0	80 60	57.15
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	80 70	56.08
7   290   0   5   9.5	70	57.41
0   310   8   4   63	70	72.61
9   510   8   4   0.5	80	68.80
10   510   4   4   0.5 11   270   4   2   63	60 60	70.22
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	60	70.22
12   510   6   2   0.5 13   310   8   2   126	60	70.87
13   510   6   2   12.0	80	70.81
14   270   4   4   0.5 15   310   8   2   63	80	75.24
15    510    6    2    0.5	60 60	71.82
10   270   0   2   12.0   17   17   270   8   2   12.6   17.0	80	52 41
17   270   6   2   12.0   12.0   18   290   6   3   95	70	53.84
10   250   6   3   95	70	51 22
20 $270$ $4$ $4$ $12.6$	70 80	52.00
21 $310$ $4$ $2$ $63$	60	50.68
21   310   4   2   0.5 22   270   4   2   126	60	51.02
22   270   10   2   12.0   1	70	50.23
23   250   10   310   8   4   126	60	52.33
25 $290$ $6$ $3$ $9.5$	70	62.62
26 290 6 3 9.5	90	63.13
27 290 6 5 9.5	70	65.66
28 270 4 4 6.3	60	67.86
29 270 8 2 6.3	80	65.68
30 290 6 3 15.8	70	69.67
31 270 4 2 12.6	80	74.64
32 310 8 2 12.6	80	70.28
33 310 4 4 6.3	60	60.12
34 270 8 2 6.3	60	56.24
35 270 4 4 12.6	60	33.11
36 290 2 3 9.5	70	75.35
37 310 8 4 6.3	60	68.32
38 290 6 3 9.5	70	78.2
39 270 4 2 6.3	80	38.64

Table 1 Experimental results of microwave calcination-sulphuric acid leaching

#### **Results and discussion**

The data in Table 1 were taken as the training samples of the neural network prediction model. In network training process, network weights and threshold of the network were obtained through 10 iterations. The training convergence curve of the neural network prediction model was shown in Figure 1. From the figure 1, it found that the convergence precision was 0.00001, which achieved the goal.





Fig.1 Iterative convergence curve of the anti-predicted model

Figure 2 showed the leaching rate of germanium value curve and the actual 3D graph for microwave heating temperature (270~350°C), liquid-solid ratio (4~10 mL/g), leaching time (1~5 h), initial concentration of sulphuric acid (6~12 mol/L)and leaching temperature (60~80°C).







Fig.2 Comparisons between actual and predicted results of Ge extraction

As seen from the Figure 2, the prediction results of the prediction model are basically consistent with the actual values. In summary, microwave heating temperature, liquid-solid ratio and initial concentration of sulphuric had great effects on the leaching rate of germanium. It could be concluded that the model had a good adaptability and accuracy. Therefore, it was feasible to use this neural network model to predict the leaching percentages of Ge from ZnO dust by microwave calcination-sulphuric acid leaching.

#### Conclusions

The neural network model was established for predicting the leaching percentages of Ge from ZnO dust through microwave calcination-sulphuric acid leaching process. Through the network training and verification for this model, it showed that the predicted values coincided well with the

experimental values. And the convergence accuracy for this model reached 0.00001. Therefore this model could be used to predict the production conditions required for different production purposes in the process of Ge recovery, which could help to reduce the groping process and the production cost.

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