





method ( ionic liquid assisted solvent evaporation process) is a possible route to synthesize  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> with high surface area and uniform mesopore size.

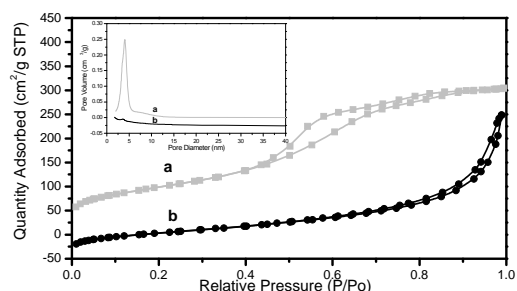


Figure 4. N<sub>2</sub> adsorption and desorption isotherms and pore size distribution (inset) for the  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> samples with [Omim]<sup>+</sup>Cl<sup>-</sup> (a) and (b) without [Omim]<sup>+</sup>Cl<sup>-</sup>

TABLE 1. TEXTURAL PROPERTY OF  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> WITH OR WITHOUT [Omim]<sup>+</sup>Cl<sup>-</sup>

Samples	Textural Property			
	BET Surface Area (m <sup>2</sup> /g)	T-polt micropore area (m <sup>2</sup> /g)	Pore Volume (cm <sup>3</sup> /g)	Average Pore Size (nm)
With [Omim] <sup>+</sup> Cl <sup>-</sup>	355.74	-0.0029	0.45	5.27
Without [Omim] <sup>+</sup> Cl <sup>-</sup>	190.83	19.84	0.46	8.27

#### IV. CONCLUSION

Here, we reported a new method (ionic liquid assisted solvent evaporation process) to synthesize  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> with high surface area and uniform mesopore size. The study shows that [Omim]<sup>+</sup>Cl<sup>-</sup> plays an important role in the formation of mesostructured  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> materials with wormlike pore structure, which might be attributed to the strong interactions between the

ionic liquid [Omim]<sup>+</sup>Cl<sup>-</sup> and reaction particles. However, the detailed mechanism is not clear and still under way nowadays. Furthermore, the method can be applied into synthesizing other inorganic materials with high surface area and uniform mesopore size, and the mesostructured  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> developed in the present work have potential applications in catalysts, optical nanodevices.

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